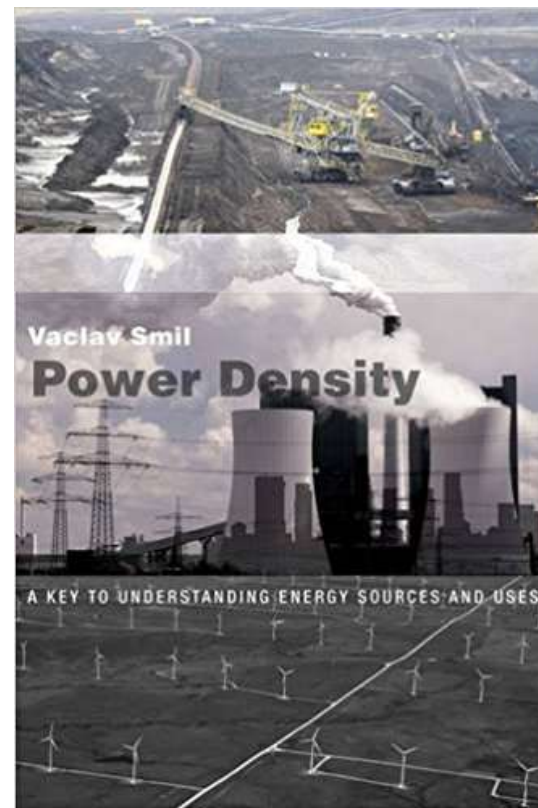
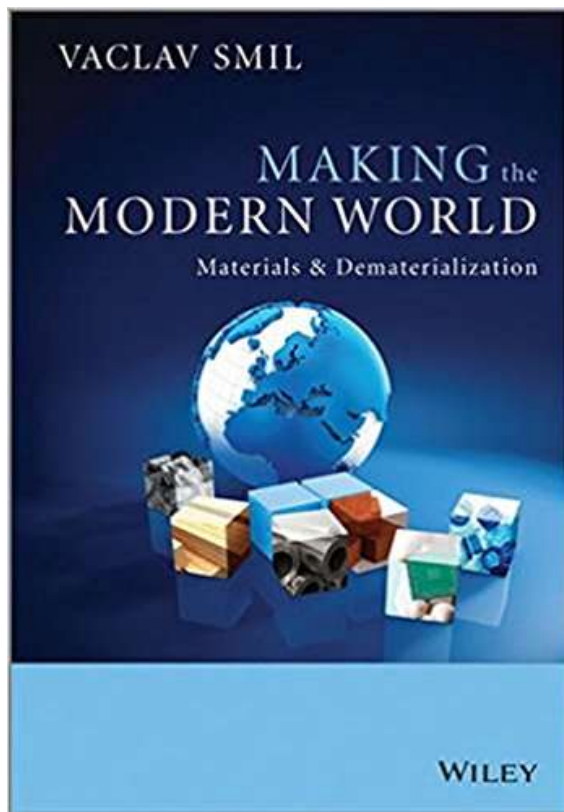
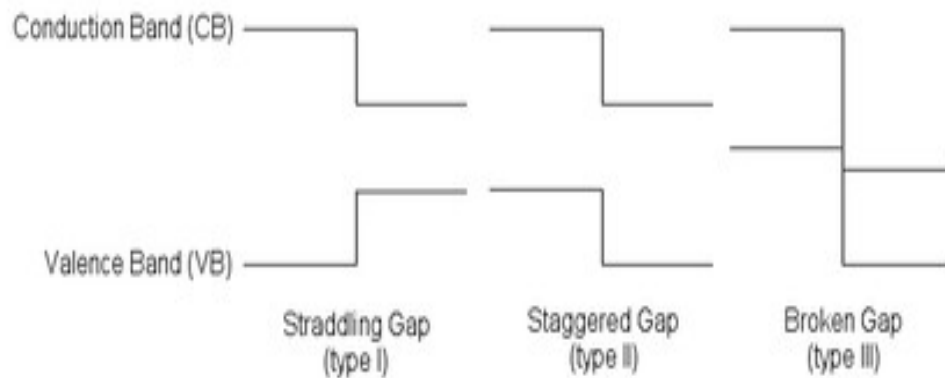


# Vaclav Smil Books

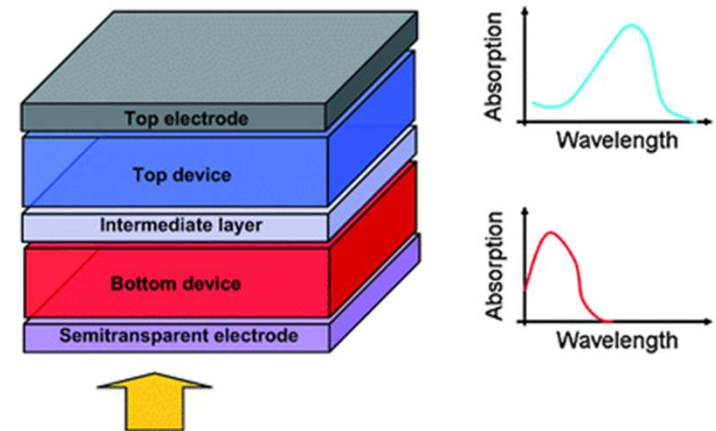


# Junctions & Tandem Solar Cells

## Junctions



## Tandem Solar Cell



# Lecture - Learning Objectives

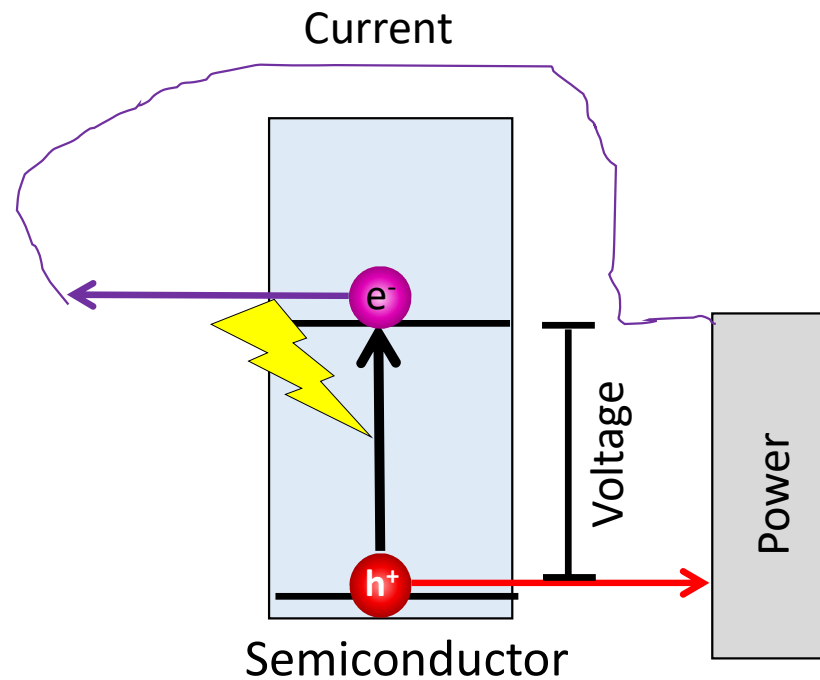


*At the end of this lecture you should:*

- Understand the following junctions:
  - p-n junction
  - p-i-n junction
  - Hetero-junction
  - Tunnel junction
  - Schottkey Barrier
- Understand how to apply various junctions to make an overall solar cell device.
- Understand basics of tandem solar cells
- Understand bi-facial solar cells

# Solar cells explained in 2 minutes

How do we get the electrons to move one way and the holes to move the other way?



# Why would an electron/hole move

- Electrons or holes move due to 1 of 2 reasons:
  - There is more  $e^-/h^+$  at one spot compared to another and they simply diffuse to a lower concentration. This is called Diffusion Current.
  - An electrochemical bias is placed across the cell, and this forces  $e^-$  and  $h^+$  to go a certain direction . This is called Drift Current.
- Mathematically the current can be written as:

$$h^+ \text{ Current} \longrightarrow J_p(r) = q \left( \overbrace{-D_p \nabla p}^{\text{Diffusion}} + \overbrace{\mu_p p F}^{\text{Drift}} \right)$$

$$e^- \text{ Current} \longrightarrow J_n(r) = q (D_n \nabla n + \mu_n n F)$$

$D_p$  = Diffusivity of holes  
 $D_n$  = Diffusivity of holes  
 $\mu_p$  = Hole mobility  
 $\mu_n$  = Electron mobility  
 $F$  = Electric Field

# p-n Junctions

# Doping a Semiconductor

i.e.  
Changing the  
Fermi level

## Periodic Table of Elements



| 1 2 3 4 5 6 7 8 9 10               |  |                                  |  |                                    |  |                                  |  |                                   |  | p-type                          | 1 n-type 15 16 17 18 |                                    |  |                                |  |                                 |  |                                 |  |                                    |  |                                 |  |                                     |  |                                 |  |                                  |  |                                 |  |                                |  |                                |  |                                 |  |                                 |  |                                 |  |                                     |  |                                 |  |                                  |  |                                     |  |                                 |  |                                  |  |                                    |  |                                   |  |                                 |  |                                  |  |                                    |  |                                    |  |                                   |  |                                    |  |                                   |  |
|------------------------------------|--|----------------------------------|--|------------------------------------|--|----------------------------------|--|-----------------------------------|--|---------------------------------|----------------------|------------------------------------|--|--------------------------------|--|---------------------------------|--|---------------------------------|--|------------------------------------|--|---------------------------------|--|-------------------------------------|--|---------------------------------|--|----------------------------------|--|---------------------------------|--|--------------------------------|--|--------------------------------|--|---------------------------------|--|---------------------------------|--|---------------------------------|--|-------------------------------------|--|---------------------------------|--|----------------------------------|--|-------------------------------------|--|---------------------------------|--|----------------------------------|--|------------------------------------|--|-----------------------------------|--|---------------------------------|--|----------------------------------|--|------------------------------------|--|------------------------------------|--|-----------------------------------|--|------------------------------------|--|-----------------------------------|--|
| 1<br>H<br>Hydrogen<br>1.00794      |  | 4<br>Be<br>Beryllium<br>9.012182 |  | 11<br>Na<br>Sodium<br>22.98976928  |  | 12<br>Mg<br>Magnesium<br>24.3050 |  | 19<br>K<br>Potassium<br>39.0983   |  | 20<br>Ca<br>Calcium<br>40.078   |                      | 21<br>Sc<br>Scandium<br>44.955912  |  | 22<br>Ti<br>Titanium<br>47.887 |  | 23<br>V<br>Vanadium<br>50.9415  |  | 24<br>Cr<br>Chromium<br>51.9961 |  | 25<br>Mn<br>Manganese<br>54.938045 |  | 26<br>Fe<br>Iron<br>55.845      |  | 27<br>Co<br>Cobalt<br>58.933195     |  | 28<br>Ni<br>Nickel<br>58.6934   |  | 29<br>Cu<br>Copper<br>63.546     |  | 30<br>Zn<br>Zinc<br>65.38       |  | 31<br>Ga<br>Gallium<br>69.723  |  | 32<br>Ge<br>Germanium<br>72.64 |  | 33<br>As<br>Arsenic<br>74.9216  |  | 34<br>Se<br>Selenium<br>78.96   |  | 35<br>Br<br>Bromine<br>79.904   |  | 36<br>Kr<br>Krypton<br>83.798       |  |                                 |  |                                  |  |                                     |  |                                 |  |                                  |  |                                    |  |                                   |  |                                 |  |                                  |  |                                    |  |                                    |  |                                   |  |                                    |  |                                   |  |
| 3<br>Li<br>Lithium<br>6.941        |  | 6<br>C<br>Carbon<br>12.0107      |  | 9<br>F<br>Fluorine<br>18.9984032   |  | 10<br>Ne<br>Neon<br>20.1797      |  | 17<br>Cl<br>Chlorine<br>35.453    |  | 18<br>Ar<br>Argon<br>39.948     |                      | 37<br>Rb<br>Rubidium<br>85.4678    |  | 38<br>Sr<br>Strontium<br>87.62 |  | 39<br>Y<br>Yttrium<br>88.90585  |  | 40<br>Zr<br>Zirconium<br>91.224 |  | 41<br>Nb<br>Niobium<br>92.90638    |  | 42<br>Mo<br>Molybdenum<br>95.94 |  | 43<br>Tc<br>Technetium<br>(97.9072) |  | 44<br>Ru<br>Ruthenium<br>101.07 |  | 45<br>Rh<br>Rhodium<br>102.90550 |  | 46<br>Pd<br>Palladium<br>106.42 |  | 47<br>Ag<br>Silver<br>107.8682 |  | 48<br>Cd<br>Cadmium<br>112.411 |  | 49<br>In<br>Indium<br>114.818   |  | 50<br>Sn<br>Tin<br>118.710      |  | 51<br>Sb<br>Antimony<br>121.757 |  | 52<br>Te<br>Tellurium<br>127.60     |  | 53<br>I<br>Iodine<br>126.90447  |  | 54<br>Xe<br>Xenon<br>131.29      |  |                                     |  |                                 |  |                                  |  |                                    |  |                                   |  |                                 |  |                                  |  |                                    |  |                                    |  |                                   |  |                                    |  |                                   |  |
| 5<br>B<br>Boron<br>10.811          |  | 8<br>O<br>Oxygen<br>15.9994      |  | 15<br>P<br>Phosphorus<br>30.973762 |  | 16<br>S<br>Sulfur<br>32.065      |  | 22<br>Fe<br>Iron<br>55.845        |  | 23<br>Co<br>Cobalt<br>58.933195 |                      | 24<br>Ni<br>Nickel<br>58.6934      |  | 25<br>Cu<br>Copper<br>63.546   |  | 26<br>Zn<br>Zinc<br>65.38       |  | 27<br>Ga<br>Gallium<br>69.723   |  | 28<br>Ge<br>Germanium<br>72.64     |  | 29<br>As<br>Arsenic<br>74.9216  |  | 30<br>Se<br>Selenium<br>78.96       |  | 31<br>Br<br>Bromine<br>79.904   |  | 32<br>Kr<br>Krypton<br>83.798    |  | 33<br>Rb<br>Rubidium<br>85.4678 |  | 34<br>Sr<br>Strontium<br>87.62 |  | 35<br>Y<br>Yttrium<br>88.90585 |  | 36<br>Zr<br>Zirconium<br>91.224 |  | 37<br>Nb<br>Niobium<br>92.90638 |  | 38<br>Mo<br>Molybdenum<br>95.94 |  | 39<br>Tc<br>Technetium<br>(97.9072) |  | 40<br>Ru<br>Ruthenium<br>101.07 |  | 41<br>Rh<br>Rhodium<br>102.90550 |  | 42<br>Pd<br>Palladium<br>106.42     |  | 43<br>Ag<br>Silver<br>107.8682  |  | 44<br>Cd<br>Cadmium<br>112.411   |  | 45<br>In<br>Indium<br>114.818      |  | 46<br>Sn<br>Tin<br>118.710        |  | 47<br>Sb<br>Antimony<br>121.757 |  | 48<br>Te<br>Tellurium<br>127.60  |  | 49<br>I<br>Iodine<br>126.90447     |  | 50<br>Xe<br>Xenon<br>131.29        |  |                                   |  |                                    |  |                                   |  |
| 13<br>Al<br>Aluminum<br>26.9815385 |  | 14<br>Si<br>Silicon<br>28.0855   |  | 15<br>P<br>Phosphorus<br>30.973762 |  | 16<br>S<br>Sulfur<br>32.065      |  | 17<br>Cl<br>Chlorine<br>35.453    |  | 18<br>Ar<br>Argon<br>39.948     |                      | 37<br>Rb<br>Rubidium<br>85.4678    |  | 38<br>Sr<br>Strontium<br>87.62 |  | 39<br>Y<br>Yttrium<br>88.90585  |  | 40<br>Zr<br>Zirconium<br>91.224 |  | 41<br>Nb<br>Niobium<br>92.90638    |  | 42<br>Mo<br>Molybdenum<br>95.94 |  | 43<br>Tc<br>Technetium<br>(97.9072) |  | 44<br>Ru<br>Ruthenium<br>101.07 |  | 45<br>Rh<br>Rhodium<br>102.90550 |  | 46<br>Pd<br>Palladium<br>106.42 |  | 47<br>Ag<br>Silver<br>107.8682 |  | 48<br>Cd<br>Cadmium<br>112.411 |  | 49<br>In<br>Indium<br>114.818   |  | 50<br>Sn<br>Tin<br>118.710      |  | 51<br>Sb<br>Antimony<br>121.757 |  | 52<br>Te<br>Tellurium<br>127.60     |  | 53<br>I<br>Iodine<br>126.90447  |  | 54<br>Xe<br>Xenon<br>131.29      |  |                                     |  |                                 |  |                                  |  |                                    |  |                                   |  |                                 |  |                                  |  |                                    |  |                                    |  |                                   |  |                                    |  |                                   |  |
| 19<br>K<br>Potassium<br>39.0983    |  | 20<br>Ca<br>Calcium<br>40.078    |  | 21<br>Sc<br>Scandium<br>44.955912  |  | 22<br>Ti<br>Titanium<br>47.887   |  | 23<br>V<br>Vanadium<br>50.9415    |  | 24<br>Cr<br>Chromium<br>51.9961 |                      | 25<br>Mn<br>Manganese<br>54.938045 |  | 26<br>Fe<br>Iron<br>55.845     |  | 27<br>Co<br>Cobalt<br>58.933195 |  | 28<br>Ni<br>Nickel<br>58.6934   |  | 29<br>Cu<br>Copper<br>63.546       |  | 30<br>Zn<br>Zinc<br>65.38       |  | 31<br>Ga<br>Gallium<br>69.723       |  | 32<br>Ge<br>Germanium<br>72.64  |  | 33<br>As<br>Arsenic<br>74.9216   |  | 34<br>Se<br>Selenium<br>78.96   |  | 35<br>Br<br>Bromine<br>79.904  |  | 36<br>Kr<br>Krypton<br>83.798  |  | 37<br>Rb<br>Rubidium<br>85.4678 |  | 38<br>Sr<br>Strontium<br>87.62  |  | 39<br>Y<br>Yttrium<br>88.90585  |  | 40<br>Zr<br>Zirconium<br>91.224     |  | 41<br>Nb<br>Niobium<br>92.90638 |  | 42<br>Mo<br>Molybdenum<br>95.94  |  | 43<br>Tc<br>Technetium<br>(97.9072) |  | 44<br>Ru<br>Ruthenium<br>101.07 |  | 45<br>Rh<br>Rhodium<br>102.90550 |  | 46<br>Pd<br>Palladium<br>106.42    |  | 47<br>Ag<br>Silver<br>107.8682    |  | 48<br>Cd<br>Cadmium<br>112.411  |  | 49<br>In<br>Indium<br>114.818    |  | 50<br>Sn<br>Tin<br>118.710         |  | 51<br>Sb<br>Antimony<br>121.757    |  | 52<br>Te<br>Tellurium<br>127.60   |  | 53<br>I<br>Iodine<br>126.90447     |  | 54<br>Xe<br>Xenon<br>131.29       |  |
| 55<br>Cs<br>Caesium<br>132.9054519 |  | 56<br>Ba<br>Barium<br>137.327    |  | 57-71<br>Lanthanoids               |  | 72<br>Hf<br>Hafnium<br>178.49    |  | 73<br>Ta<br>Tantalum<br>180.94788 |  | 74<br>W<br>Tungsten<br>183.84   |                      | 75<br>Re<br>Rhenium<br>186.207     |  | 76<br>Os<br>Osmium<br>190.23   |  | 77<br>Ir<br>Iridium<br>192.222  |  | 78<br>Pt<br>Platinum<br>195.084 |  | 79<br>Au<br>Gold<br>196.966569     |  | 80<br>Hg<br>Mercury<br>200.59   |  | 81<br>Tl<br>Thallium<br>204.3833    |  | 82<br>Pb<br>Lead<br>207.2       |  | 83<br>Bi<br>Bismuth<br>208.98040 |  | 84<br>Po<br>Polonium<br>(209)   |  | 85<br>At<br>Astatine<br>(210)  |  | 86<br>Rn<br>Radon<br>(222)     |  | 87<br>Fr<br>Francium<br>(223)   |  | 88<br>Ra<br>Radium<br>(226)     |  | 89-103<br>Actinoids             |  | 104<br>Rf<br>Rutherfordium<br>(261) |  | 105<br>Db<br>Dubnium<br>(262)   |  | 106<br>Sg<br>Seaborgium<br>(266) |  | 107<br>Bh<br>Bohrium<br>(264)       |  | 108<br>Hs<br>Hassium<br>(277)   |  | 109<br>Mt<br>Meitnerium<br>(268) |  | 110<br>Ds<br>Darmstadtium<br>(271) |  | 111<br>Rg<br>Roentgenium<br>(272) |  | 112<br>Uub<br>Ununbium<br>(285) |  | 113<br>Uut<br>Ununtrium<br>(284) |  | 114<br>Uuq<br>Ununquadium<br>(289) |  | 115<br>Uup<br>Ununpentium<br>(288) |  | 116<br>Uuh<br>Ununhexium<br>(292) |  | 117<br>Uus<br>Ununseptium<br>(294) |  | 118<br>Uuo<br>Ununoctium<br>(294) |  |

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

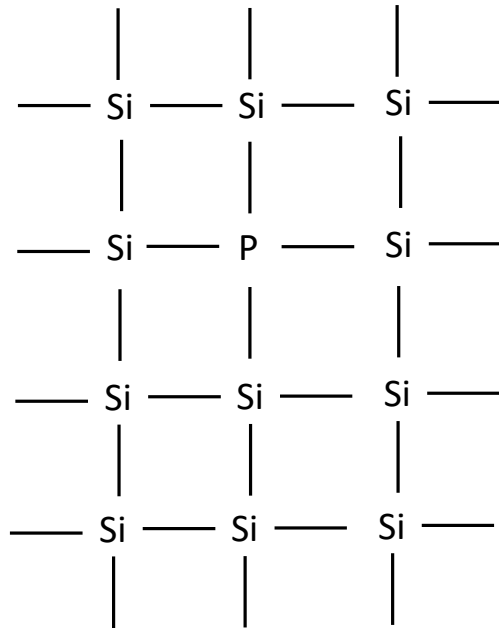
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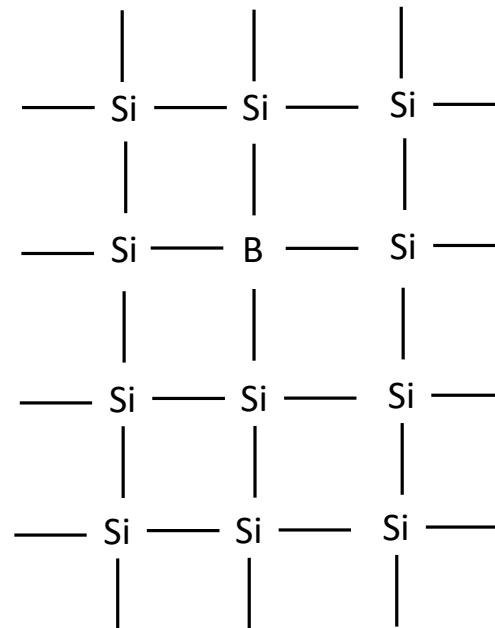
|                                   |                                  |                                       |                                  |                                 |                                |                                 |                                  |                                  |                                   |                                  |                               |                                   |                                  |                                  |
|-----------------------------------|----------------------------------|---------------------------------------|----------------------------------|---------------------------------|--------------------------------|---------------------------------|----------------------------------|----------------------------------|-----------------------------------|----------------------------------|-------------------------------|-----------------------------------|----------------------------------|----------------------------------|
| 57<br>La<br>Lanthanum<br>138.9047 | 58<br>Ce<br>Cerium<br>140.116    | 59<br>Pr<br>Praseodymium<br>140.90768 | 60<br>Nd<br>Neodymium<br>144.242 | 61<br>Pm<br>Promethium<br>(145) | 62<br>Sm<br>Samarium<br>150.36 | 63<br>Eu<br>Europium<br>151.964 | 64<br>Gd<br>Gadolinium<br>157.25 | 65<br>Tb<br>Terbium<br>158.92535 | 66<br>Dy<br>Dysprosium<br>162.500 | 67<br>Ho<br>Holmium<br>164.93032 | 68<br>Er<br>Erbium<br>167.259 | 69<br>Tm<br>Thulium<br>168.93421  | 70<br>Yb<br>Ytterbium<br>173.054 | 71<br>Lu<br>Lutetium<br>174.9668 |
| 89<br>Ac<br>Actinium<br>(227)     | 90<br>Th<br>Thorium<br>232.03806 | 91<br>Pa<br>Protactinium<br>231.03688 | 92<br>U<br>Uranium<br>238.02891  | 93<br>Np<br>Neptunium<br>(237)  | 94<br>Pu<br>Plutonium<br>(244) | 95<br>Am<br>Americium<br>(243)  | 96<br>Cm<br>Curium<br>(247)      | 97<br>Bk<br>Berkelium<br>(247)   | 98<br>Cf<br>Californium<br>(251)  | 99<br>Es<br>Einsteinium<br>(252) | 100<br>Fm<br>Fermium<br>(257) | 101<br>Md<br>Mendelevium<br>(258) | 102<br>No<br>Nobelium<br>(259)   | 103<br>Lr<br>Lawrencium<br>(262) |

# How to get an n-type or p-type material

- Here are the most common ways to make Si n-type or p-type.



n-type



p-type

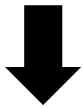
- Here are the most common ways to make Si n-type or p-type.



# Separating the charge

- Creating a p-n junction allows us to create a built in electrochemical bias.
- At the interface charge transfers between n-side and p-side due to diffusion until it builds enough potential to make current = 0.

$$J_n(r) = q(D_n \nabla n + \mu_n n F)$$

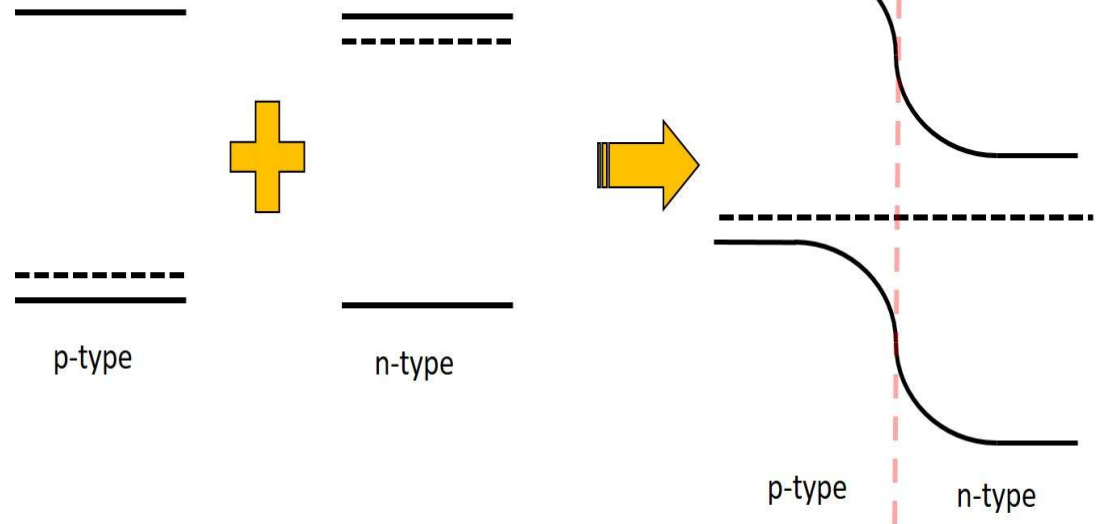


$$0 = q(D_n \nabla n + \mu_n n F)$$

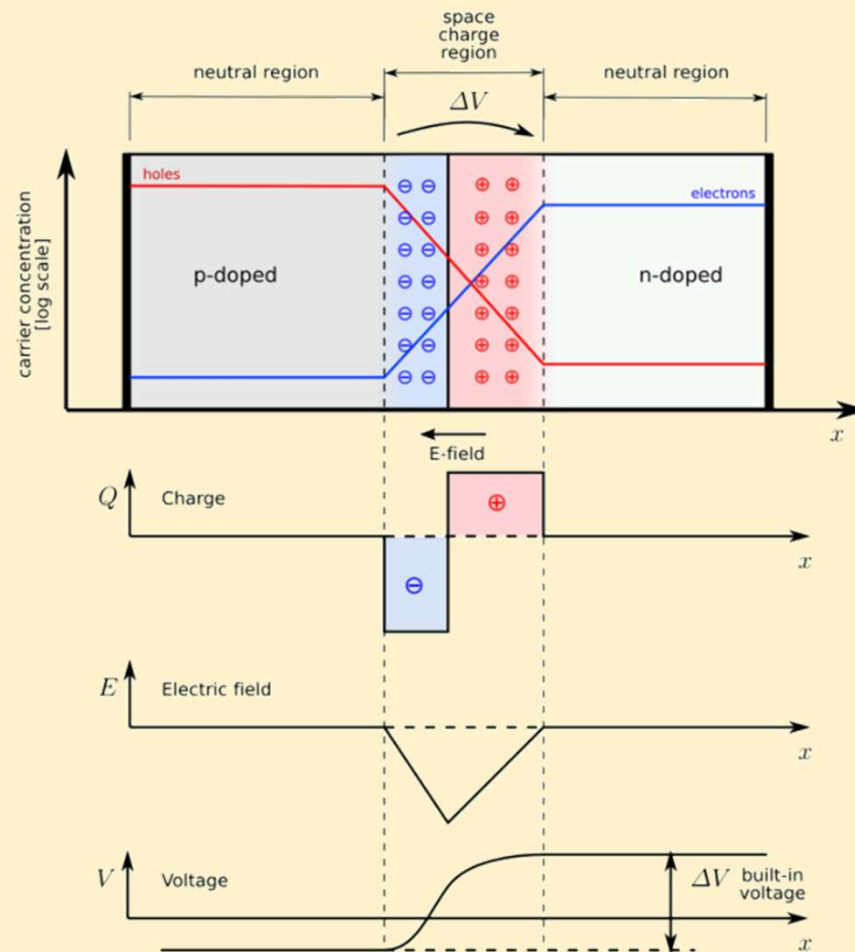


$$F_{n-side} = \frac{-qD_n \nabla n}{\mu_n n}$$

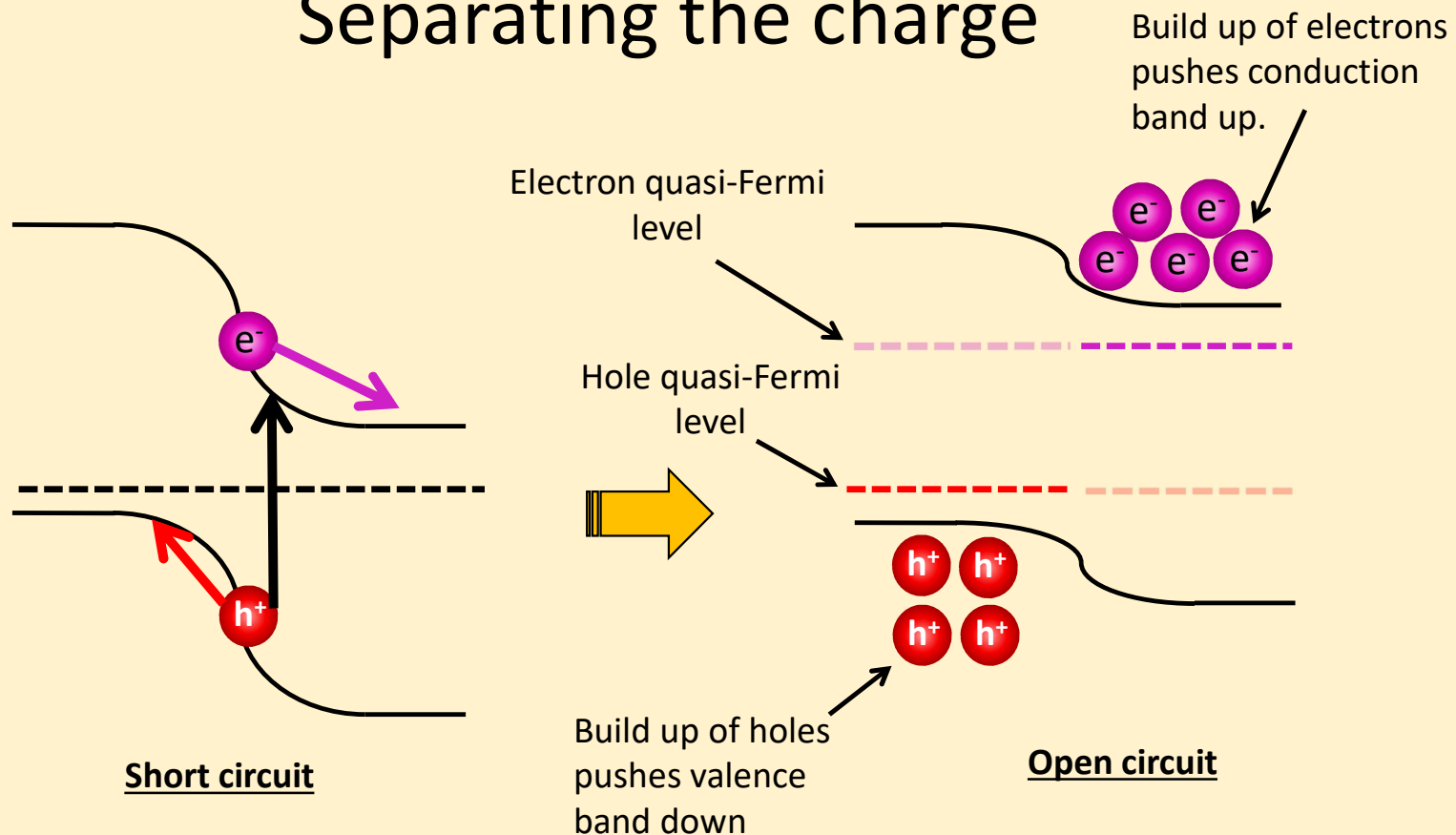
Gradient at p-n junction



# Breaking down the p-n junction

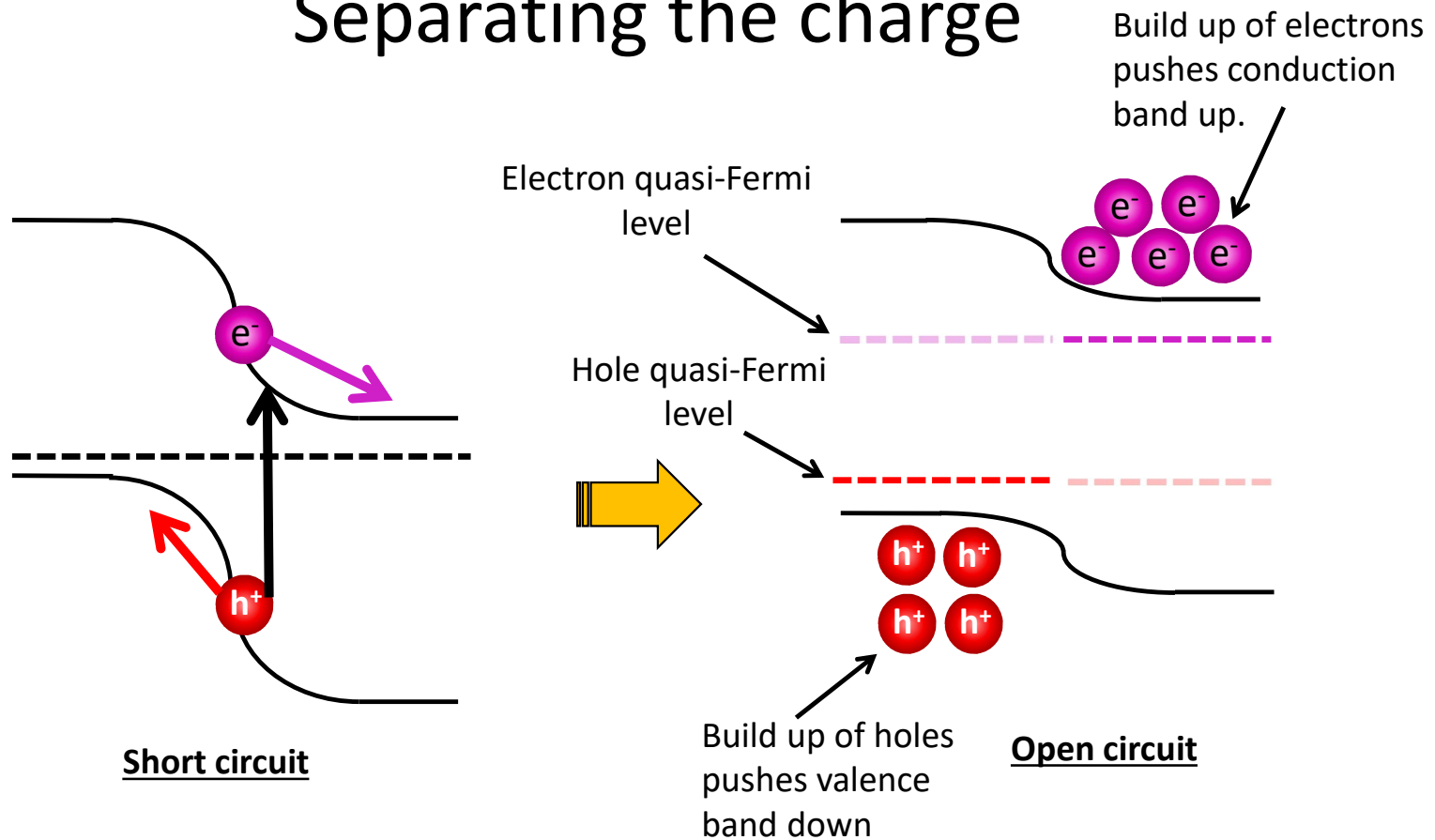


# Separating the charge



- Will the band-bending thickness always be equally between the p and n side?

# Separating the charge



# Doping Profiles

- Basically the high dopant density material has a stronger pull than the lower doped material.
- By using Poisson's equation to distribute the charge, we can determine the depletion width.

Depletion width  
(band bending  
thickness )

$$x_p = \left( \frac{N_D}{N_A} \frac{2\epsilon\epsilon_0(V_{bias} - V_{photo})}{q(N_D + N_A)} \right)^{1/2}$$

$$x_n = \left( \frac{N_A}{N_D} \frac{2\epsilon\epsilon_0(V_{bias} - V_{photo})}{q(N_D + N_A)} \right)^{1/2}$$

$N_A$  = Acceptor (p-type) density

$N_D$  = Donor (n-type) density

$\epsilon$  = permittivity of semiconductor

$\epsilon_0$  = permittivity in vacuum

$V_{bias}$  = Total band-bending voltage (difference in p-n Fermi levels)

$V_{photo}$  = Photo voltage

$q$  = elementary charge

# Doping Profiles

- The voltage drop on a certain side is inversely proportional to the doping density.

$$V_{p-ty} = V_{Bias} \times \frac{N_D}{N_D + N_A}$$

$$V_{n-ty} = V_{Bias} \times \frac{N_A}{N_D + N_A}$$

$N_A$  = Acceptor (p-type) density  
 $N_D$  = Donor (n-type) density

- $V_{bias}$  is the potential difference in Fermi levels between the n-type and p-type semiconductor.
- For more detailed description of these equations see [Malizia, et al, JMC-A, 2014](#).

## Concept Check

In a p-n junction solar cell, a p-type side has a dopant density of  $10^{16} \text{ cm}^{-3}$  and an n-type side has a dopant density of  $10^{20} \text{ cm}^{-3}$ . This means:

- a) The p-type has a larger voltage drop across it and a longer depletion layer.
- b) The p-type has a larger voltage drop across it and a shorter depletion layer.
- c) The n-type has a larger voltage drop across it and a longer depletion layer.
- d) The n-type has a larger voltage drop across it and a shorter depletion layer.

## Concept Check

In a p-n junction solar cell, a p-type side has a dopant density of  $10^{16} \text{ cm}^{-3}$  and an n-type side has a dopant density of  $10^{20} \text{ cm}^{-3}$ . This means:

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- c) The n-type has a larger voltage drop across it and a longer depletion layer.
- d) The n-type has a larger voltage drop across it and a shorter depletion layer.

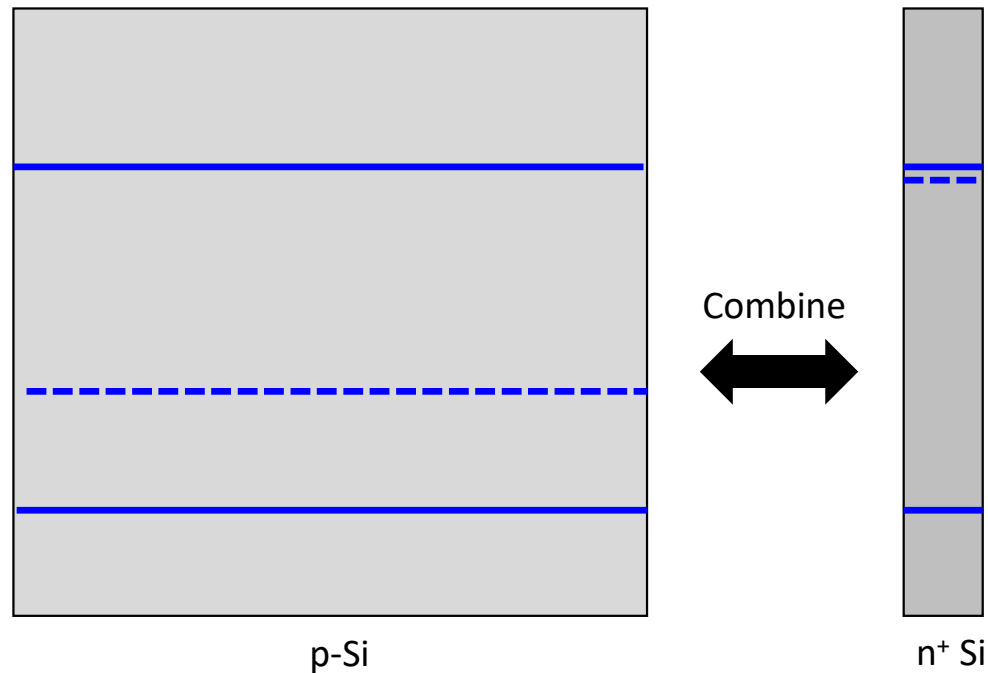


# Fundamental issue

- We need doping to get n-type or p-type and allow for band bending.
- As the difference in dopant density increases, the band bending increases.
- However dopants are basically contaminants, thus they are recombination sites.
- How do we mitigate this issue?

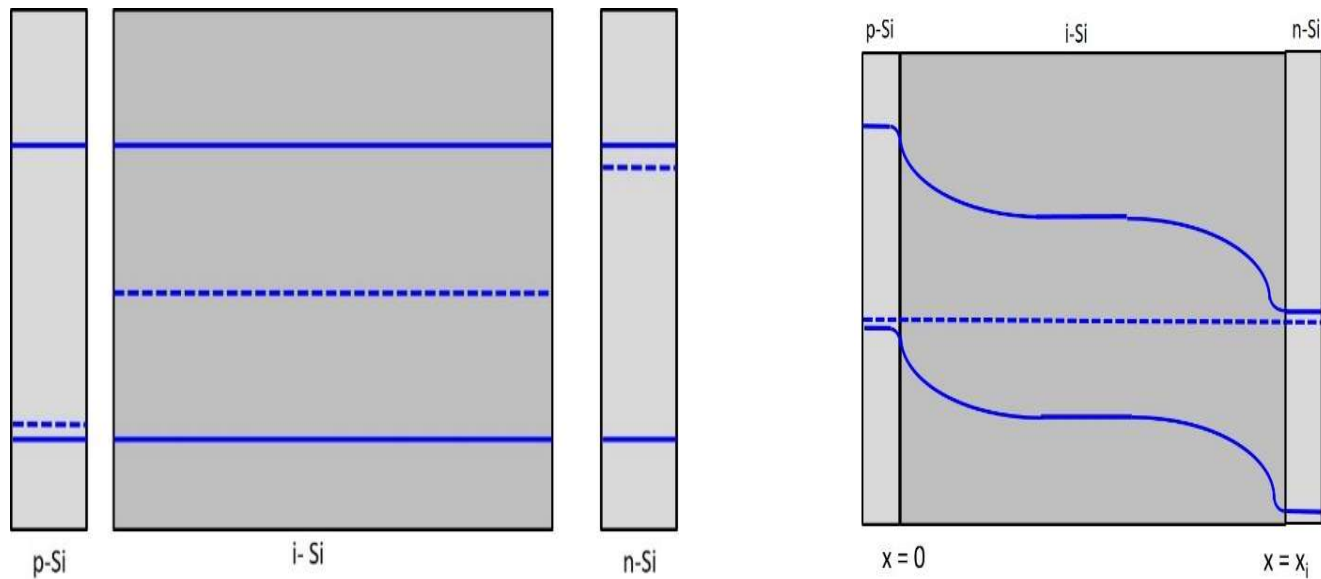
## Creating a very highly doped layer

- By having one material highly doped, and the other low doped, we induce large bandbending in the low-doped material.



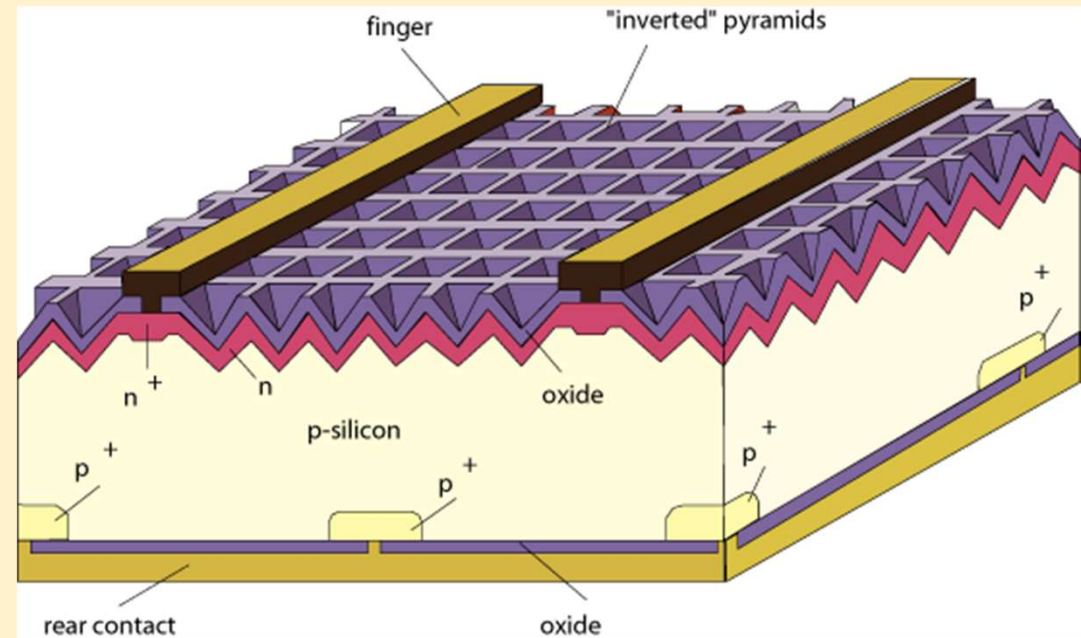
# p-i-n junction solar cell

- This approach maximizes depletion layer thickness by having band bending from both sides.
- Almost all solar cells use at least some form of a p-i-n junction.



# State of the art from 1999 - 2013

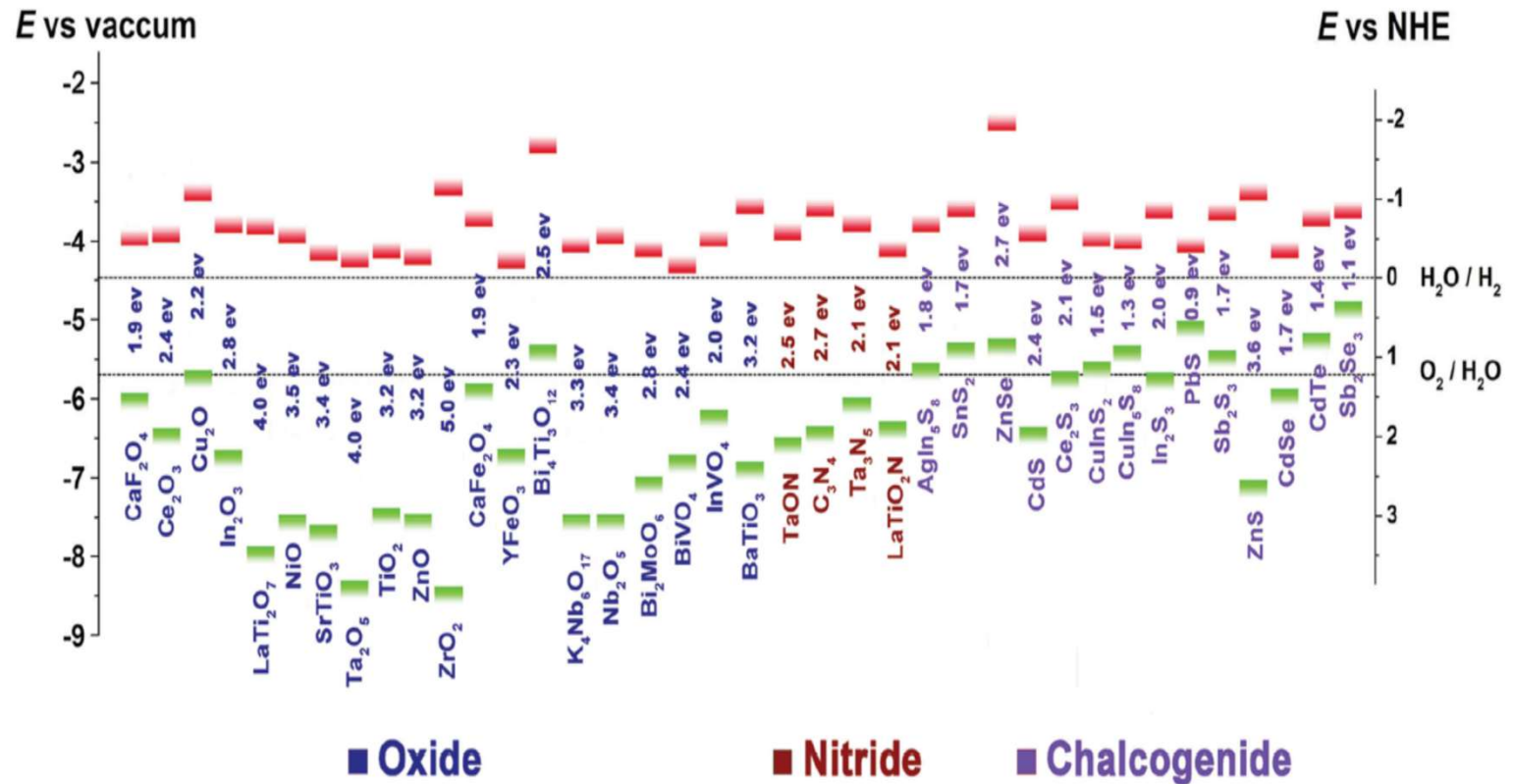
- Creating small point contacts allows for minimal surface recombination.
- Doping both sides prevents charge carriers from going to the wrong contact.
- Creating small point contacts allows for minimal surface recombination.
- This approach gave a record efficiency of 24.7%.
- What is the biggest failure of this device?



# Different approaches to junctions

# Heterojunction

- Heterojunctions is a junction between 2 different materials
- The band alignment is a function of the materials electron affinity.



# p-n Hetero-junction

- The band mismatch can create a discontinuity at the interface
- Besides this discontinuity, the junction acts like a normal p-n junction.

## Advantages

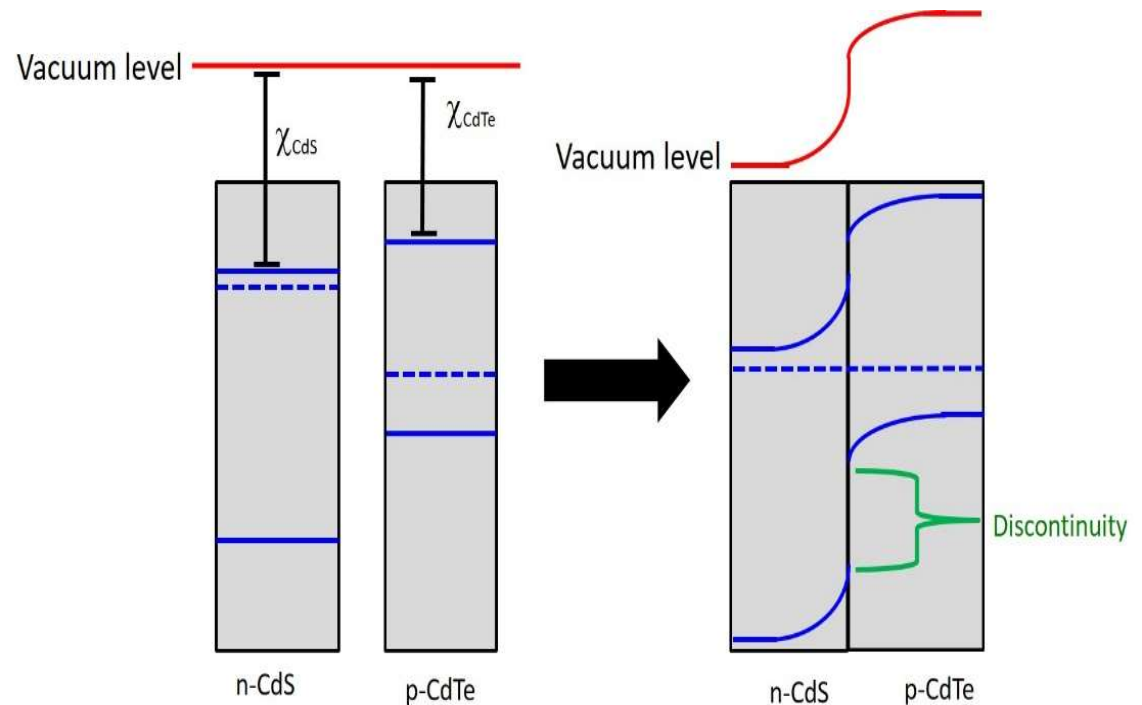
Different band gaps

Can create extreme  $V_{\text{Bias}}$

## Disadvantages

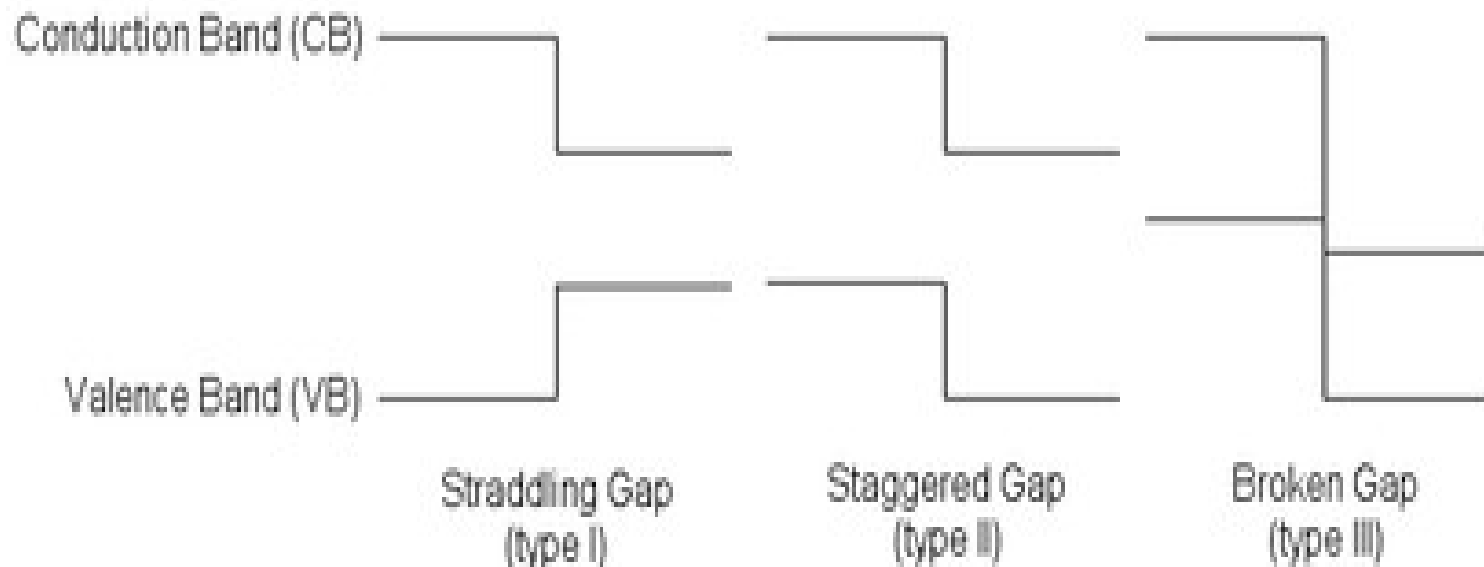
Heavy recombination at the interface

Materials need to be compatible



# Types of hetero-junctions

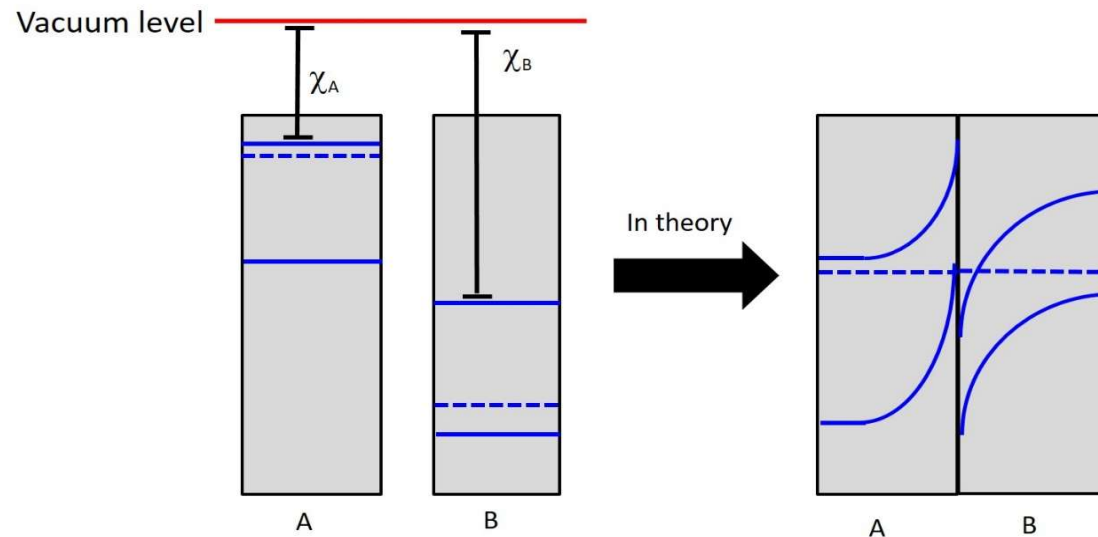
- We classify heterojunctions into 3 types
- Type 2 is almost always the type we are interested in.
- Type 1 is great in that it can prevent surface recombination and acts as a wall





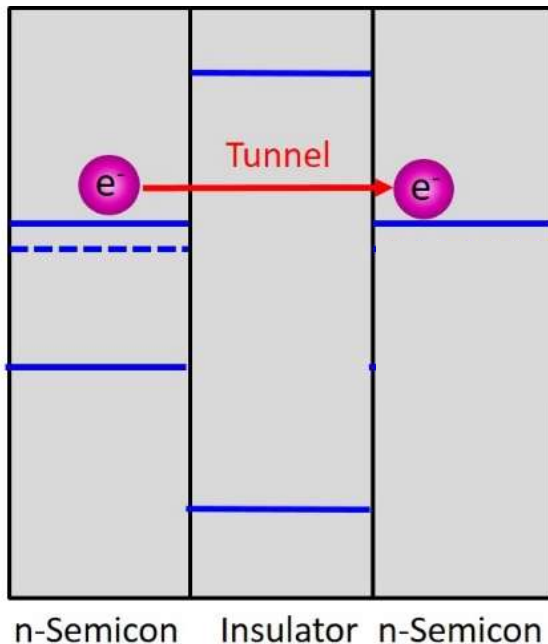
# Type 3 junction

- Type 3 is interesting to look at fundamentally.
- This design puts electrons in the conduction band of the p-type and holes in the valence band of n-type even without light.
- There is a barrier at the interface to prevent electrons and holes, thus this approach will fail.



# Type 1- Tunnel Junction

- Quantum mechanics says that our electrons can tunnel through a small barrier.
- Our electrons can only tunnel through a short distance (typically 1-5 nm).



Equations for current density ( $J_t$ ) and tunneling probability ( $T_t$ )

$$J_t = -N_C v_{th,e} q T_t$$

$$T_t = \exp \left[ -\frac{4}{3} \Delta x \sqrt{\frac{2m_{eff} q \phi_b}{\hbar^2}} \right]$$

$T_t$  = tunneling probability

$x$  = tunneling distance

$M_{eff}$  = effective mass of  $e^-$  or  $h^+$

$\hbar$  = plancks constant divided by 2 Pi

$\phi_b$  = barrier height

$N_C$  = effective density of sates

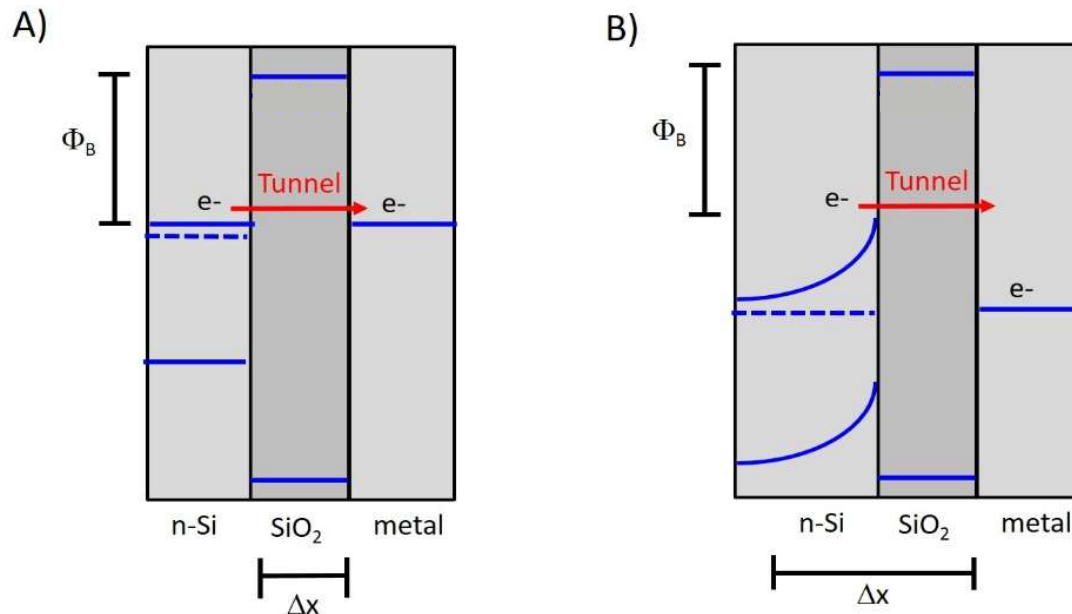
$V_{th}$  = thermal velocity

$q$  = elemenray charge

$J_t$  = tunneling current

# Tunneling with band bending

- Metals have different 'Fermi levels with respect to vacuum', which is known as 'work functions.'
- If work function is not the same as the conduction (or valence) band, this will create band bending. Thus we need to tunnel through depletion width + tunnel barrier.

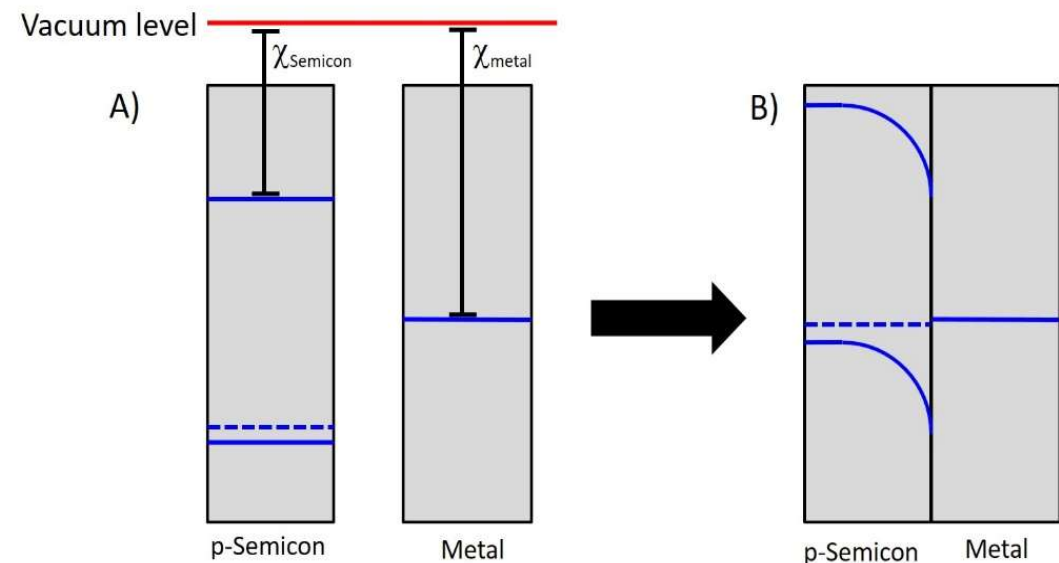


$$T_t = \exp \left[ -\frac{4}{3} \Delta x \sqrt{\frac{2m_{eff}q\Phi_b}{\hbar^2}} \right]$$

$$J_t = -N_C v_{th,e-} q T_t$$

# Schottky Barrier

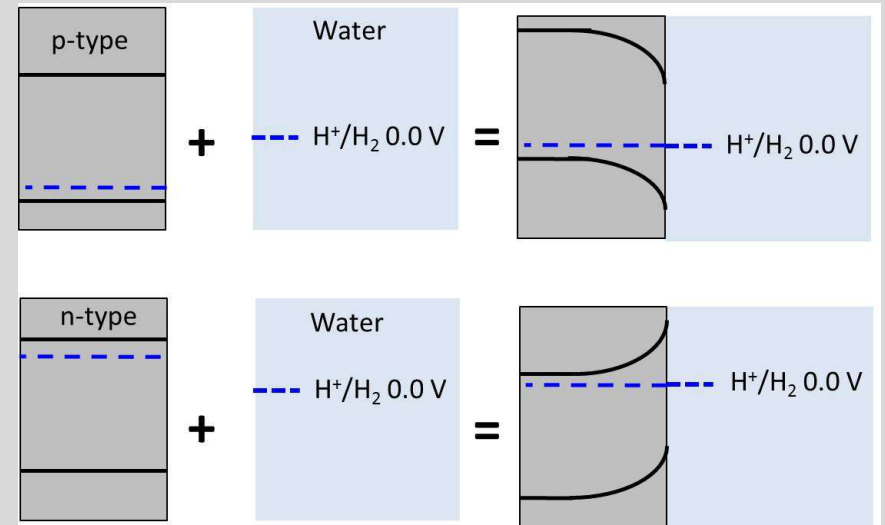
- A Schottky barrier occurs from a metal-semiconductor junction.
- A metal can simply be thought of as a infinitely doped semiconductor.
- We try to avoid these since they rarely line up with a valence or conduction band, and create a lot of defects
- If the semiconductor is highly doped, then tunnelling completely dominates. This then acts like an '*ohmic contact*'.



# Water-Semiconductor junction

- Water can be used as a junction for semiconductors.
- It is not a good junction, but it can work.

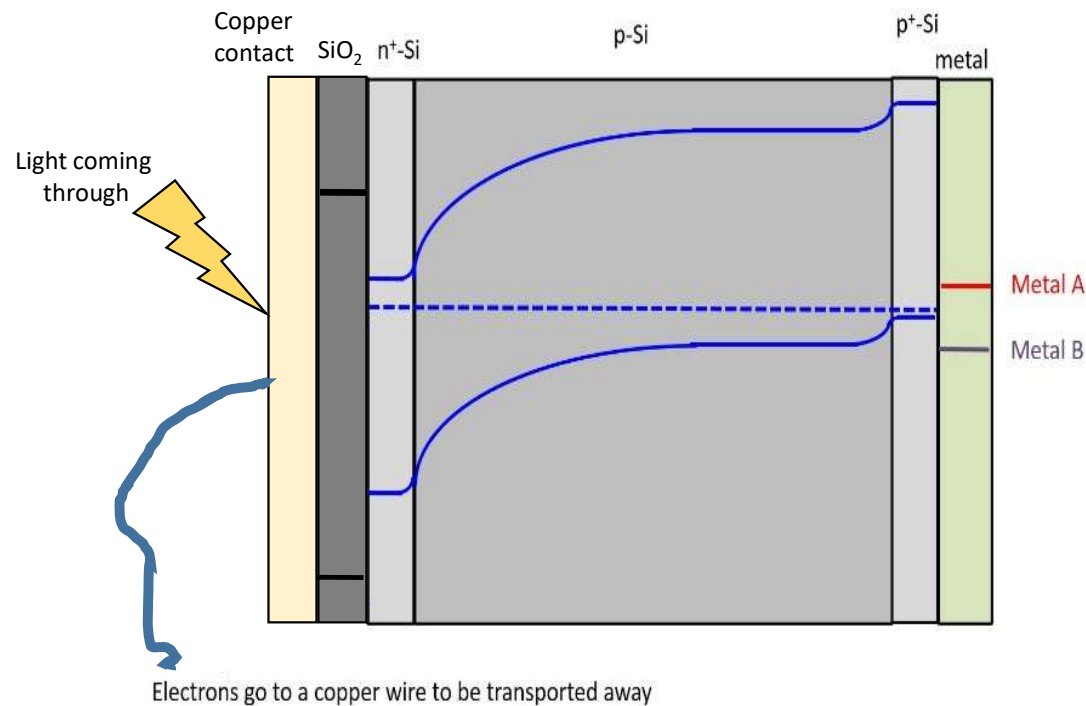
- What happens to the electron when it goes into water?
- Answer- It chemically reduces a species in the water. If there is no species it will reduce water to  $H_2$  gas and  $OH^-$ .



- The 'Energy Storage' part will focus just on the electron going from metal into solution. There is still an electric field and a potential

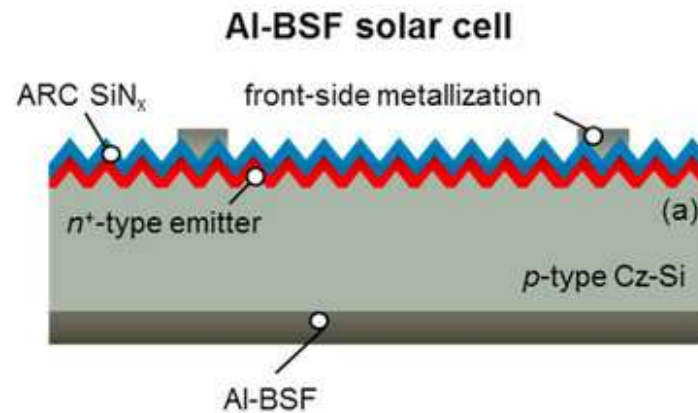
# Complete solar cell

- This is a full solar cell with all junction included.
- All solar cells don't have all type of junctions, and people are still trying to find the optimal strategy.



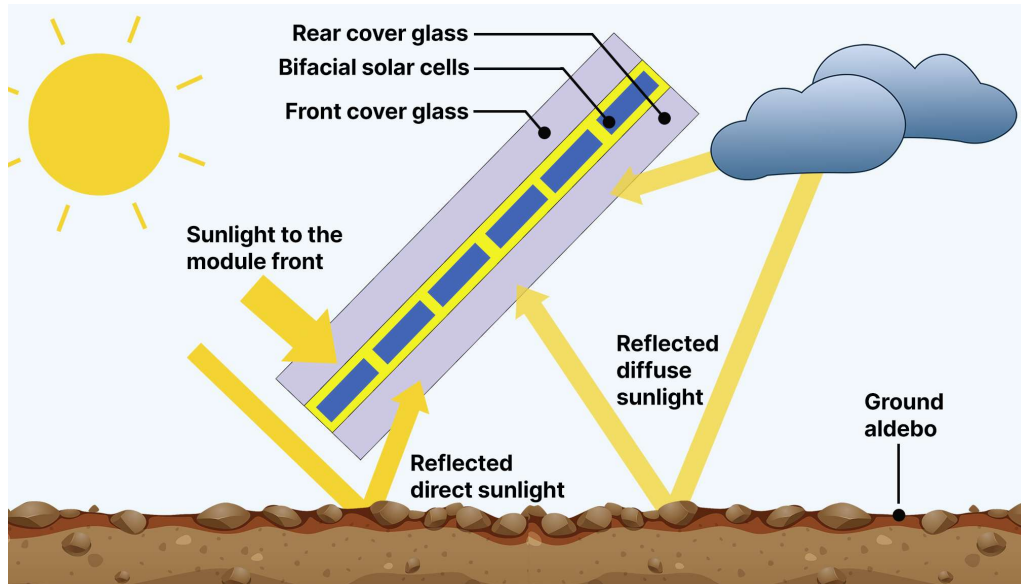
# Commercial Si Solar Cells

- The aluminum back surface field (BSF) had >80% of the market until 2015
- Relatively simple design, with no minimal technical challenges
- Efficiency of near 20%



DOI: [10.1016/j.egypro.2017.09.308](https://doi.org/10.1016/j.egypro.2017.09.308)

# Bifacial Solar Cells



<https://www.solarreviews.com/blog/bifacial-solar-panels>



<https://www.longi.com/en/news/6933/>



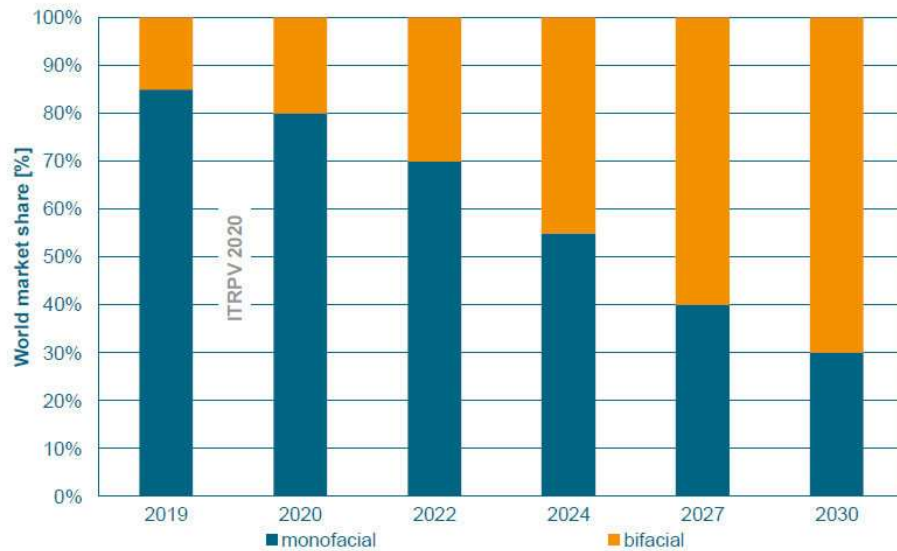
# Bifacial Solar Cells

- The idea is simple- let reflected light get absorbed from the back side of the solar cell
- A single solar cell is used, meaning the back needs to be transparent.
- This has potential for solar cell fields, but not roof top solar cell.
- The actual efficiency increases is very hard to accurately calculate, but it believed to be 10-20% of an improvement

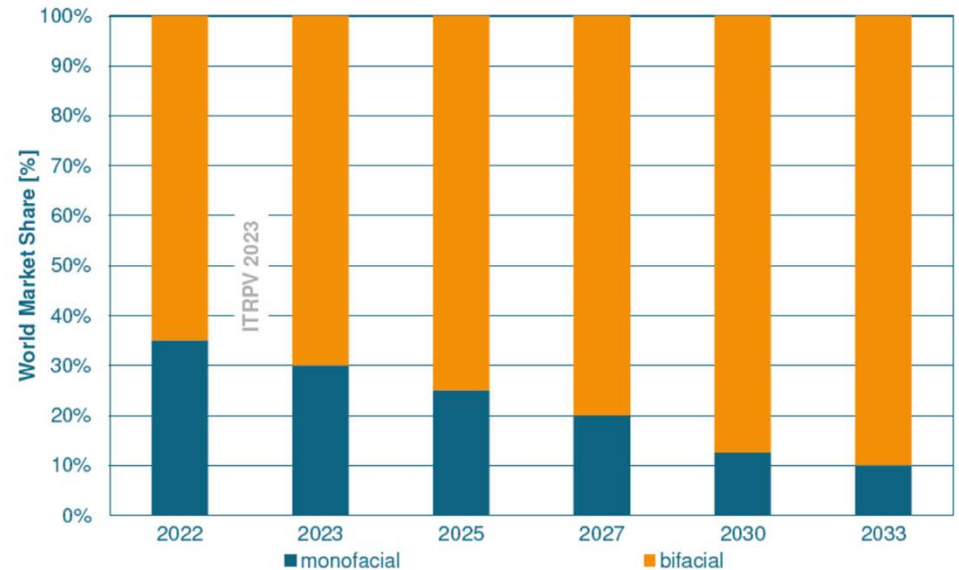
# Market Share: monofacial vs bifacial

- Below is percentage of bifacial solar cells in world market

Prediction in 2020



Prediction in 2023

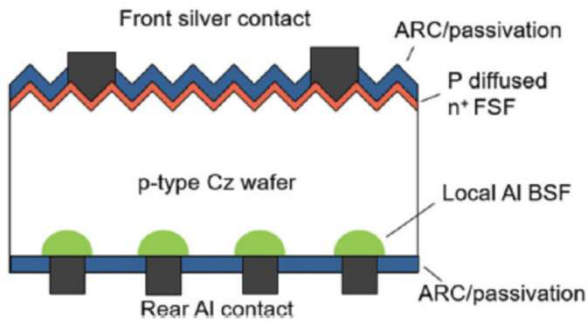


# Commercial Si Solar Cells

BSF= Back surface field  
 ARC= Anti-reflective coating  
 Emitter = highly doped

## PERC

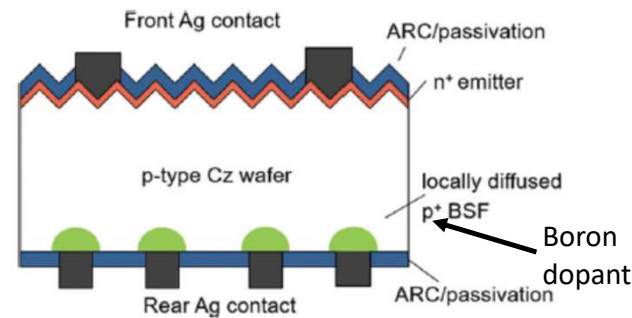
*Passivated emitter rear contact*



Efficiency : 24.5%  
 Bifaciality: ~ 80%

## PERT

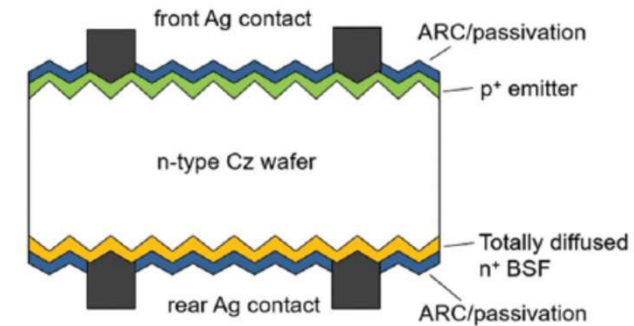
*Passivated emitter rear locally diffused*



Efficiency : 24.5%  
 Bifaciality: ~ 89%

## PERL

*Passivated emitter rear locally diffused*



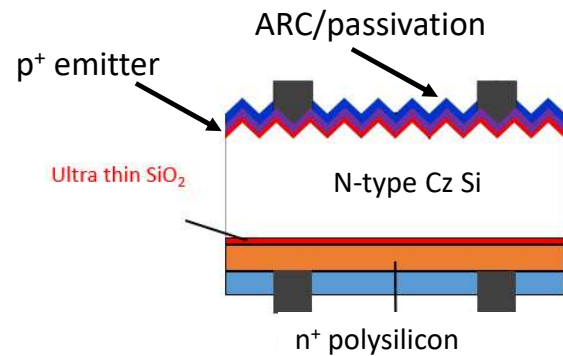
Efficiency: 24.7 %  
 Bifaciality: ~ 85%

# Commercial Solar Cells

BSF= Back surface field  
ARC= Anti-reflective coating

## TOPCon

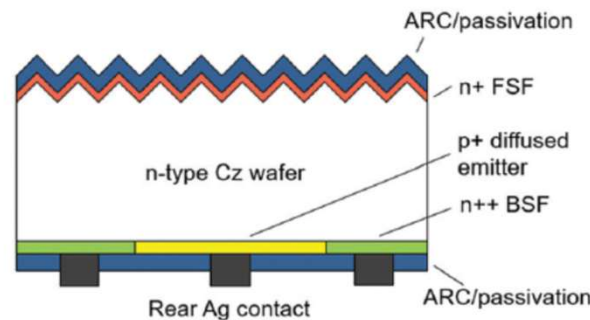
*Passivated emitter rear locally diffused*



Efficiency: 26.1 %  
Bifaciality: ~ 85%

## IBC

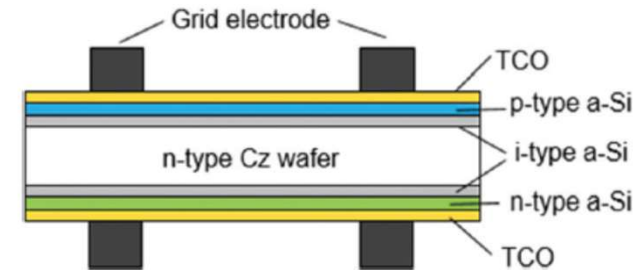
*Interdigitated back contact*



Efficiency: 26.7%  
Bifaciality: ~ 75%

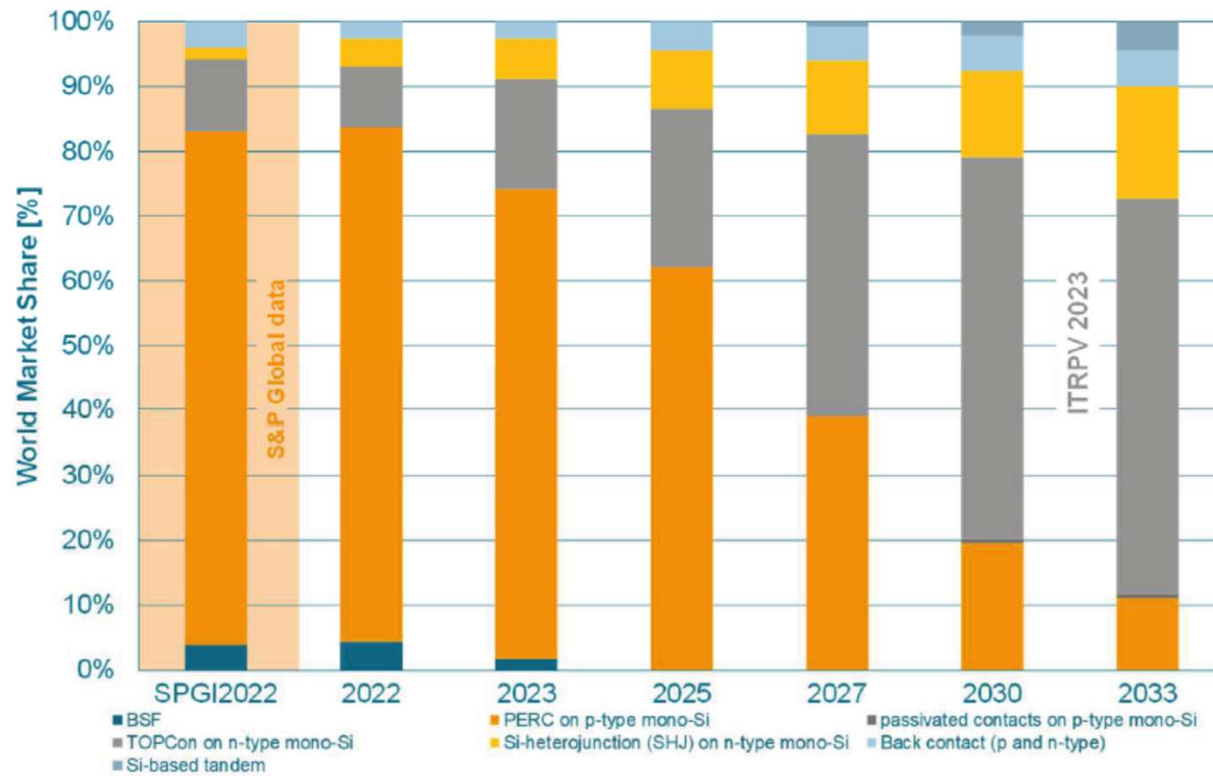
## HIT


*Heterojunction with intrinsic thin-layer*



Efficiency (front): 26.5 %  
Bifaciality: ~ 95%

# Market Share



-  Si-Tandem
-  IBC
-  HIT
-  TOPCon
-  PERC
-  Aluminum BSF

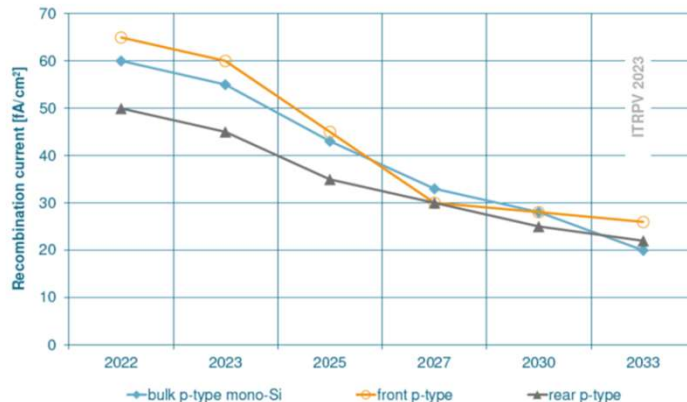
# Losses throughout a solar cell

- We can put losses at different spots in terms of dark saturation current,  $J_0$ .

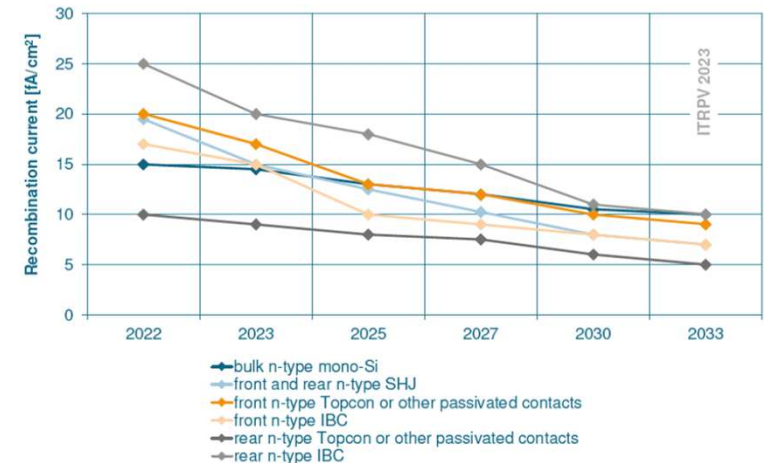
$$J_{net} = J_{sc} - J_0 \left[ \exp\left(\frac{qV}{mkT}\right) - 1 \right]$$

- While the front and rear losses are comparable to the bulk, the bulk has many  $e^-$  and  $h^+$ , where the front and back mostly have either  $e^-$  or  $h^+$ . Both  $e^-$  and  $h^+$  are needed for recombination.

Dark saturation current density  
p-type material



Dark saturation current density  
n-type material



# Electrical Contacts

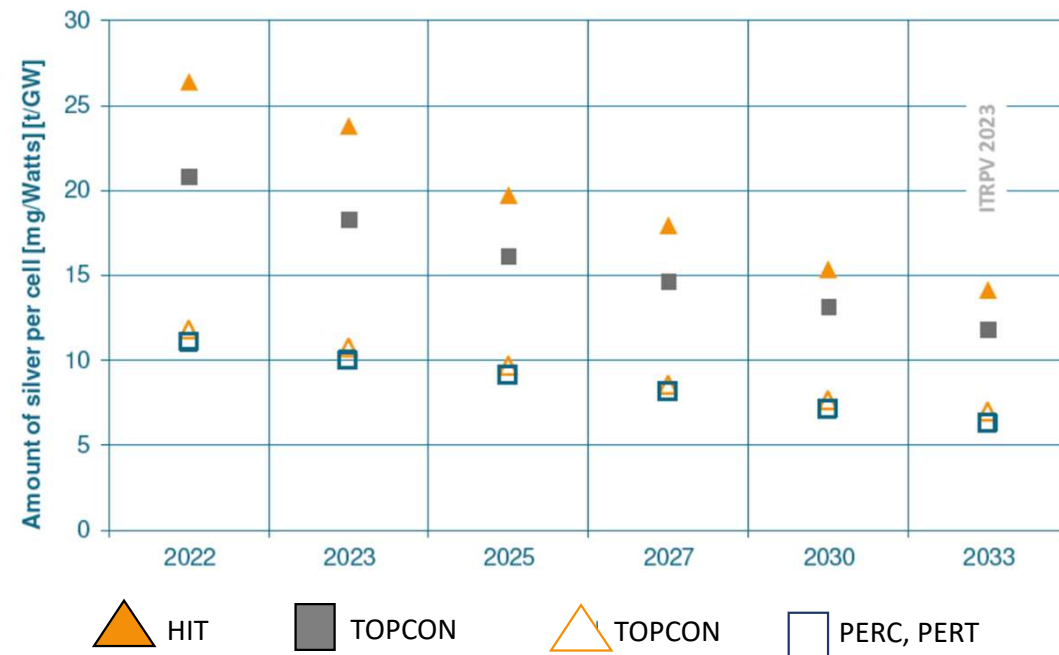
- Ag's conductivity means it is used for connections.

- The price of Ag could effect which solar type is optimal.

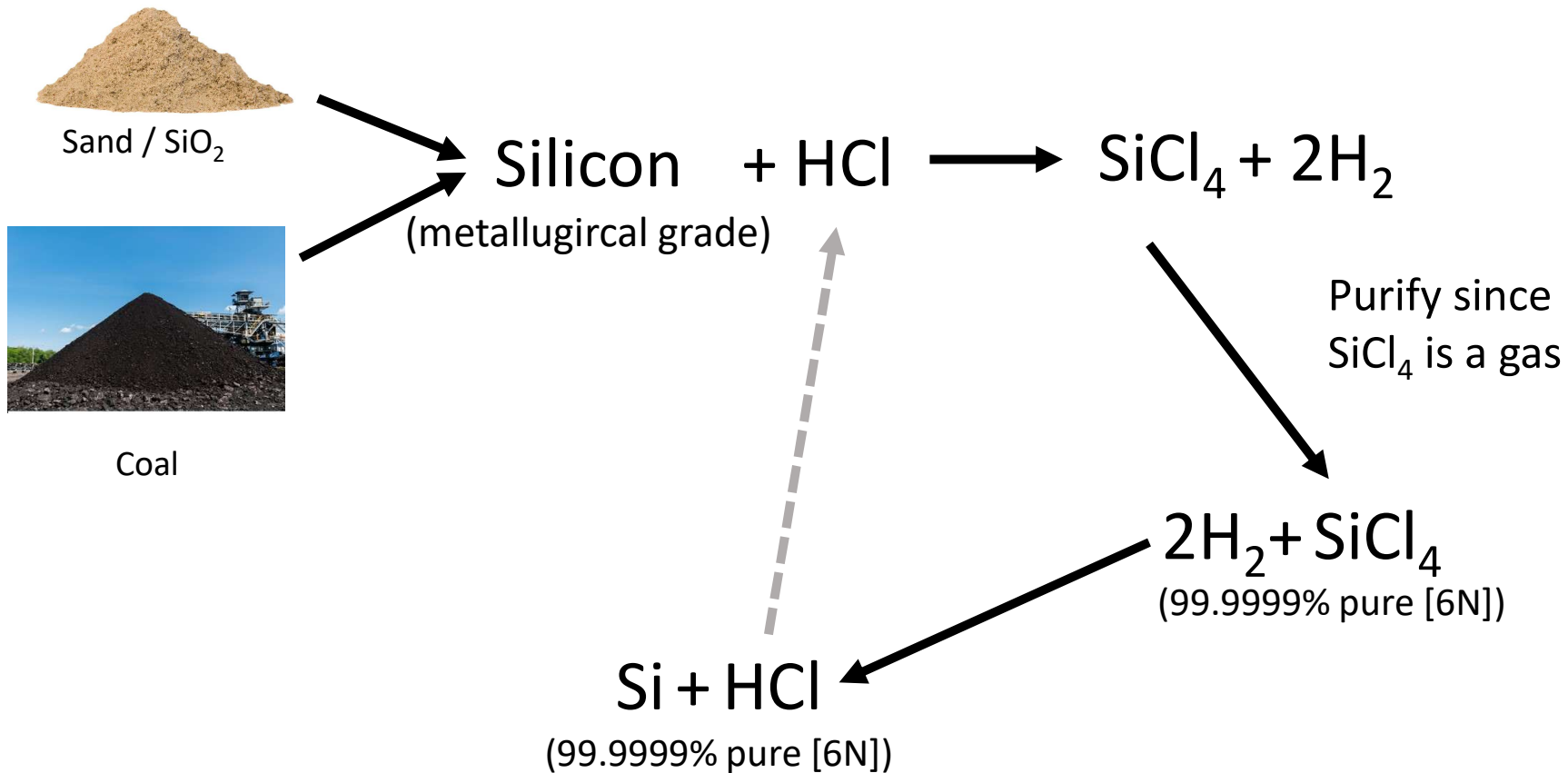
| Metal    | Conductivity, $\sigma$ ( S/cm) |
|----------|--------------------------------|
| Silver   | $6.3 \times 10^5$              |
| Gold     | $6.0 \times 10^5$              |
| Copper   | $4.5 \times 10^5$              |
| Aluminum | $3.5 \times 10^5$              |

## Trend for remaining silver for metallization per cell (front + rear side)

(Values for M6, M10, and G12 cell size, average)



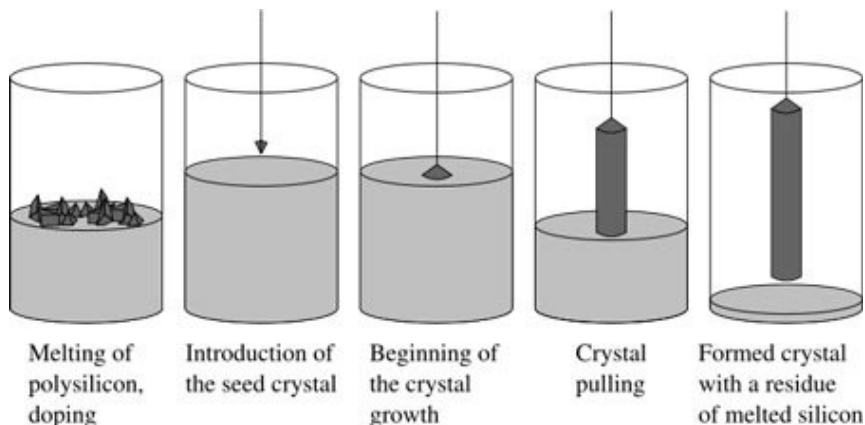
# From Sand to Silicon





# Czochralski (CZ) Si Production

- The Czochralski process of making single crystalline Si is the most popular (75% of all Si)
- The silicon is heated to 1500 °C to melt it.
- A seed crystal pulls up the Si at a rate of 25 mm per hour.
- The wafer size are anywhere from 100mm diameter to 450mm in diameter and 1-2 meters long.
- They have oxygen dopants of  $10^{18} \text{ cm}^{-3}$



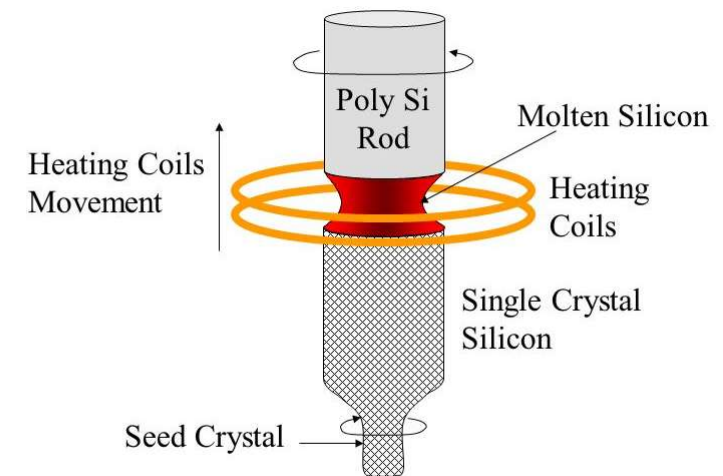
Silicon ingot

# Float Zone Si production

- Float-Zone is a more pure way to make single crystalline Si
- A poly-crystalline Si is set on a seed of crystalline Si.
- The Si sidewalls does not touch anything so it is 'floating'
- A inductive heating coil heats the Silicon and melts a small portion of it
- Once it cools, it settles into a single crystal.
- The melt also concentrates and remove contaminants.
- Typically this Si is too expensive for solar cells.



# TOPSiL

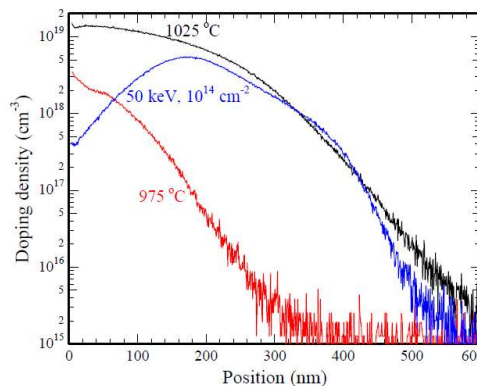


Hong Xiao, Ph. D.

[www2.austin.cc.tx.us/HongXiao/Book.htm](http://www2.austin.cc.tx.us/HongXiao/Book.htm)

# Fabrication of p-n junction

- Creating a highly doped  $p^+$  layer (or  $n^+$  layer is very easy) for Si. (the '+' means highly doped)
- For  $p^+$  dopant put in a boron based gas at elevated temperatures ( $\sim 1000$  C) for about 10 minutes.
- For  $n^+$  dopants use a phosphorous gas (such as  $\text{POCl}_3$ ).
- The boron (or phosphide) reaches a saturation concentration, thus allowing for high doping.

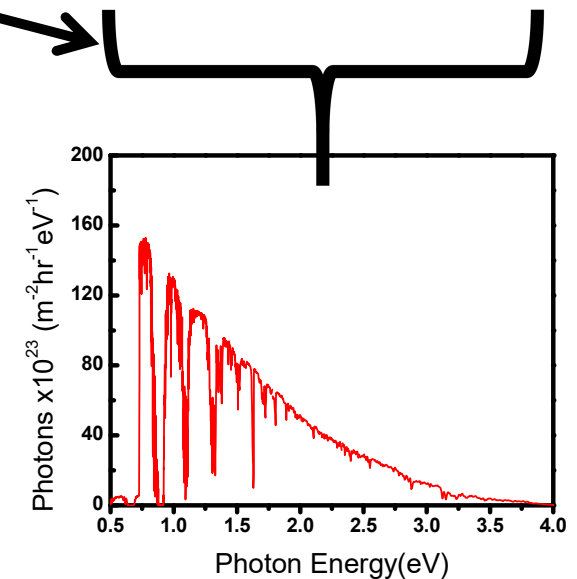
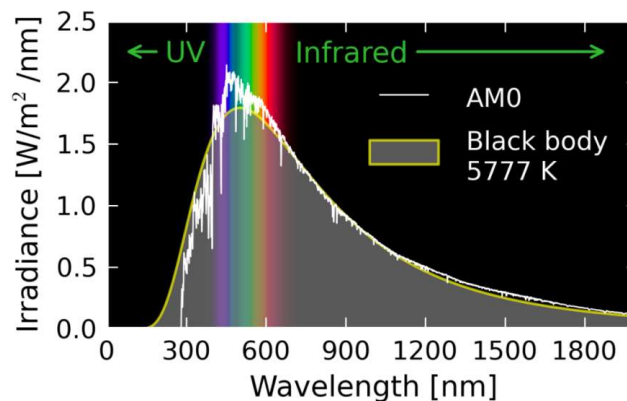
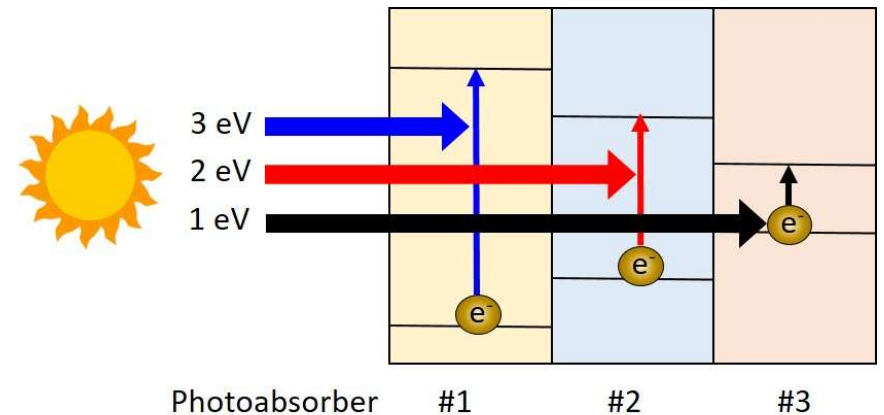


**Fig. S6.** Dopant profiles (SIMS) of the high (1025 °C) and low (975 °C) doped  $p^+$ -Si layer in the 100 nm  $\text{TiO}_2/5$  nm  $\text{Ti}/p^+n$  Si composites. Also shown is the doping profile used for calibration of the SIMS profiles ( $1 \times 10^{14} \text{ cm}^{-2} \text{ }^{11}\text{B}$  implanted at 50 keV).

# Tandem Solar Cells

# Tandem Solar Cells- Basic Principles

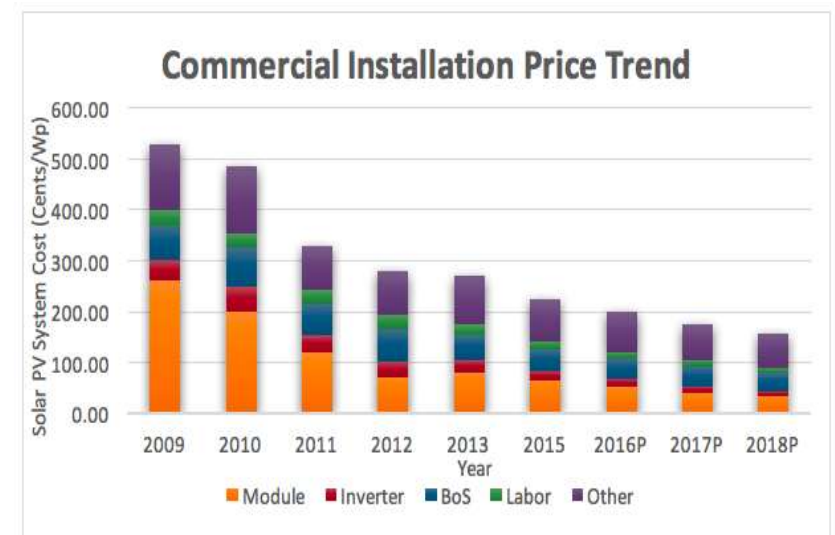
- An optimal solar cell, would have a bandgap material for each wavelength from the sun
- For economic reasons, we are interested in 2 or 3 photoabsorbers for a tandem solar cell



# Why tandems are becoming relevant



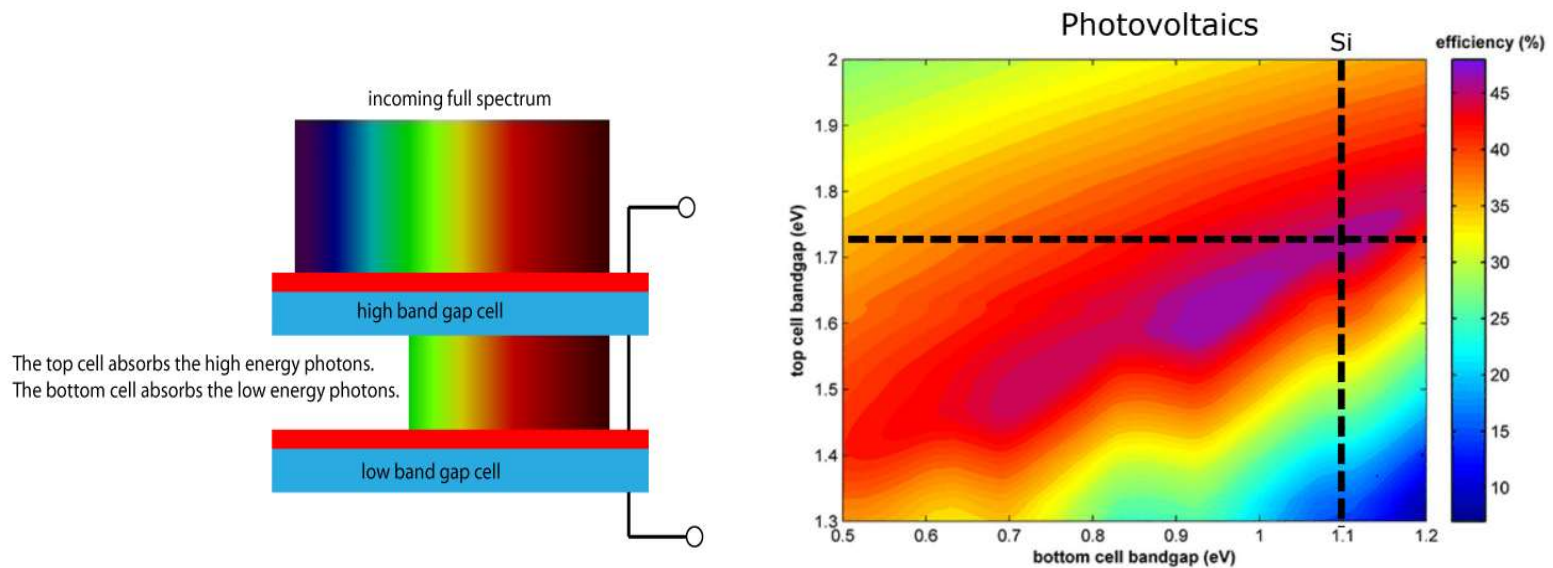
- Since the beginning, the actual solar cell was the largest cost factor for producing solar based electricity.
- While we have been able to greatly reduce solar cell costs, the balance of plant costs have been hard to reduce.
- Now balance of plant costs are the largest costs for solar cells.
- Balance of plant includes
  - Inverters
  - Permitting
  - Steel frames to hold the solar cells
  - Labor
- Tandems allow us to have the same balance of plant, but higher efficiency.



<http://www.ezysolare.com>

# How efficient can we get realistically ?

- Here is a figure relating optimal efficiencies of a 2 photoabsorber device as a function of their band gap.



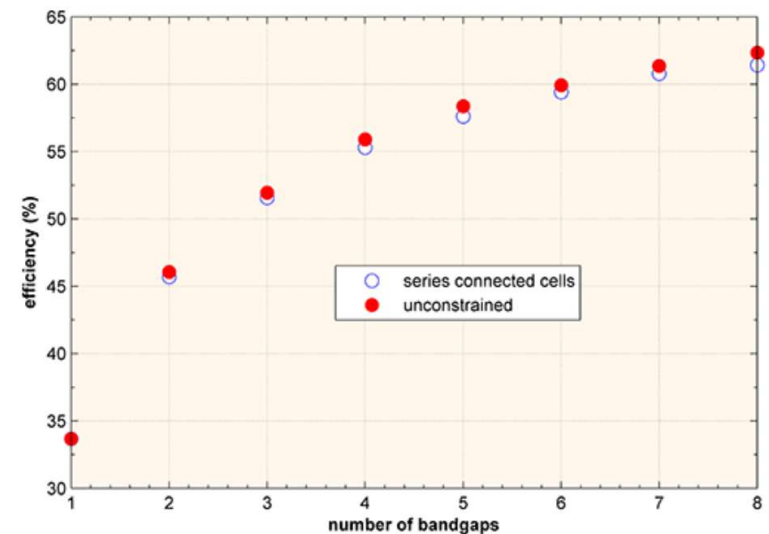
Taken from: <https://www.pveducation.org/pvcdrom/tandem-cells>

# Optimal band gaps for tandem devices

- The graph below is for optimized solar cells.

| # of cells in tandem device | Bandgap #1 | Bandgap #2 | Bandgap #3 | Bandgap #4 |
|-----------------------------|------------|------------|------------|------------|
| #1                          | 1.3        |            |            |            |
| #2                          | 1.9        | 1.0        |            |            |
| #3                          | 2.3        | 1.4        | 0.8        |            |
| #4                          | 2.6        | 1.8        | 1.2        | 0.8        |

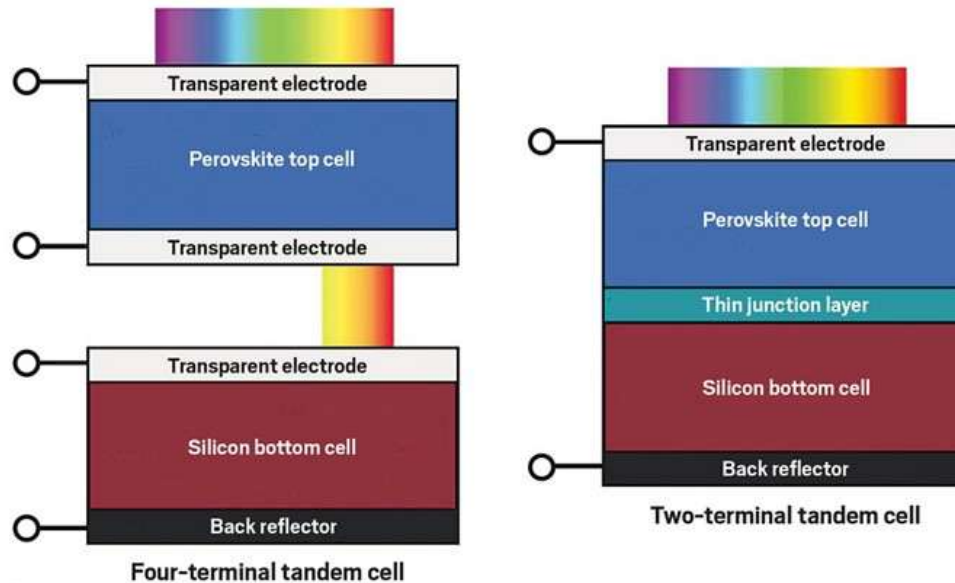
[Marti et al., Solar Energy Materials and Solar Cells 43 \(1996\) 203-222](#)



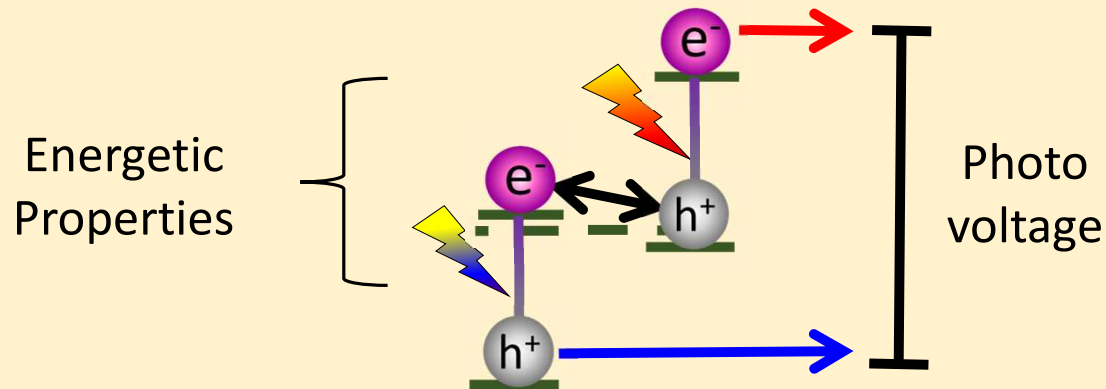
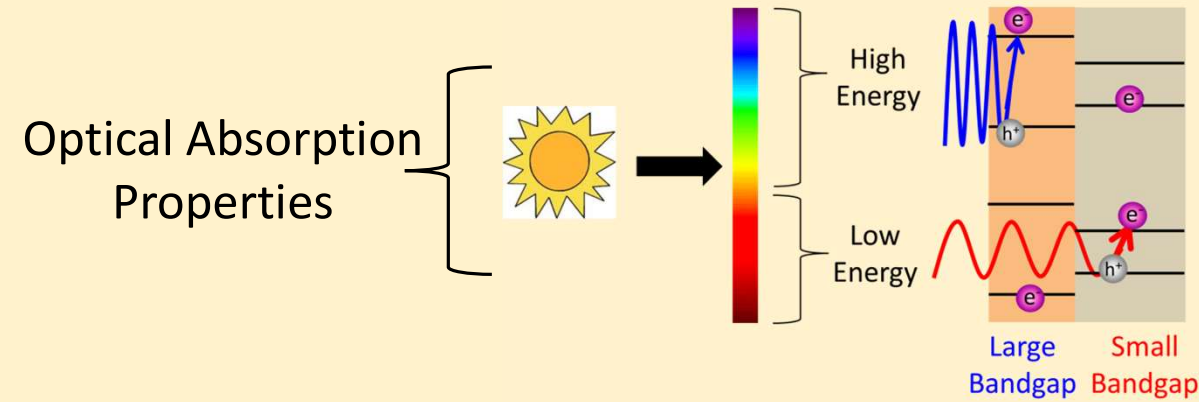


# Tandem Solar Cells- Basic Principles

- There are different approaches to making this work.
- The four terminal is simpler to construct.
- The two terminal is theoretically the better way to do it.



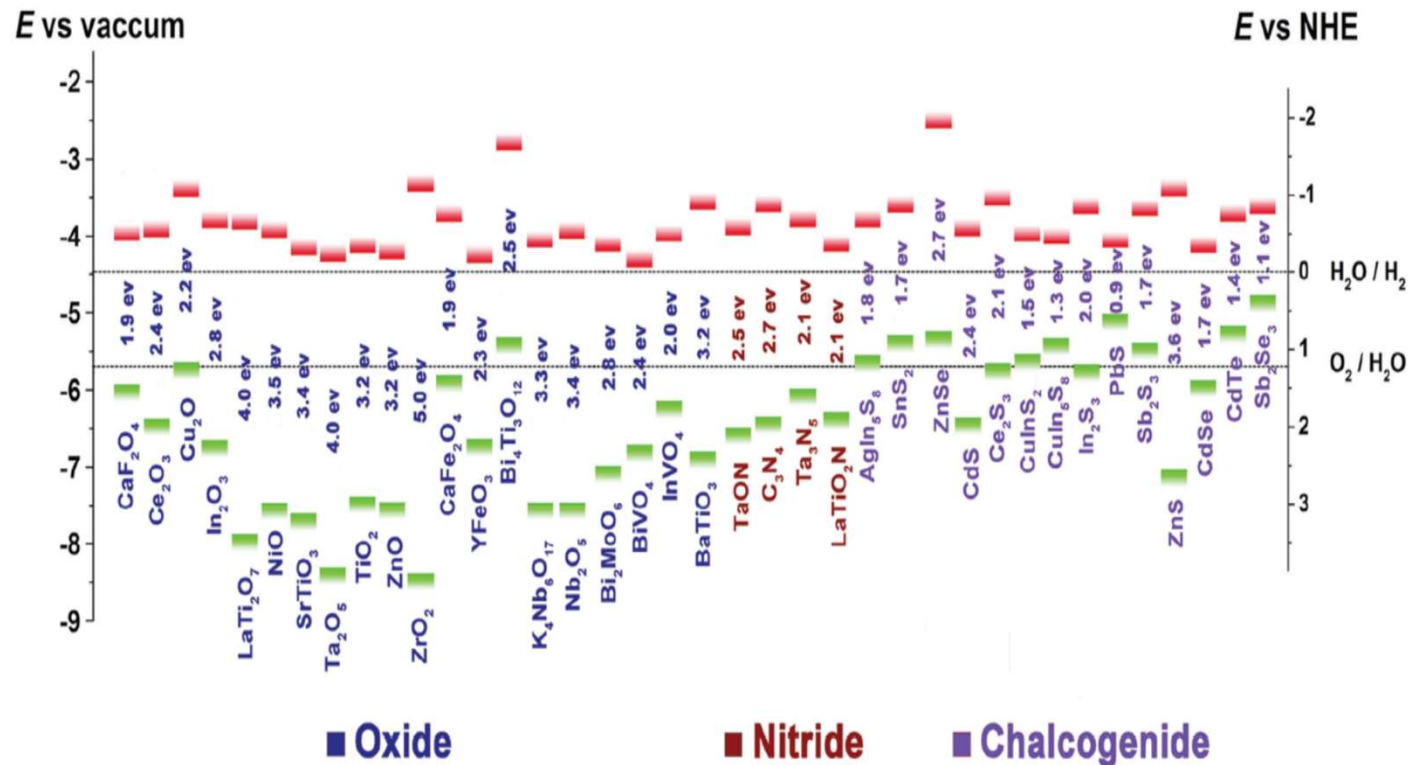
# Tandem voltage in 2 terminal approach



- Half the current, but more than double the photovoltage
- Do the energy levels automatically line up?

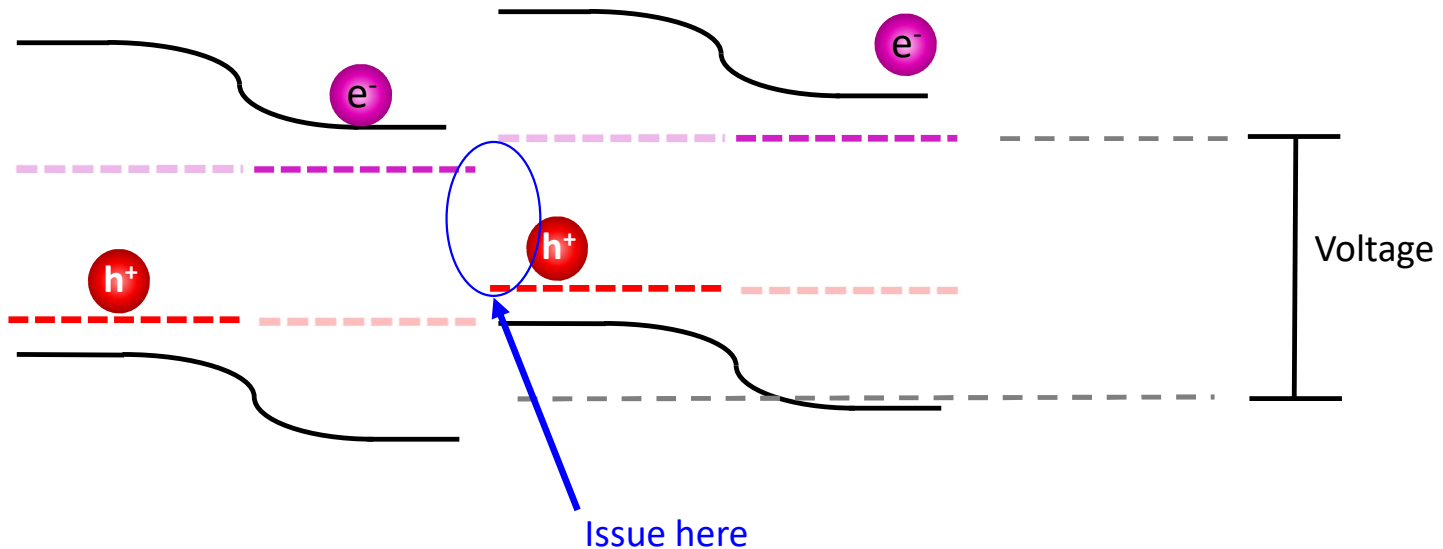
# Band position alignment

- It is going to be difficult to find good band alignment between 2 photoabsorbers (that also have optimal band gaps, and semiconductor physics properties)



# Theory

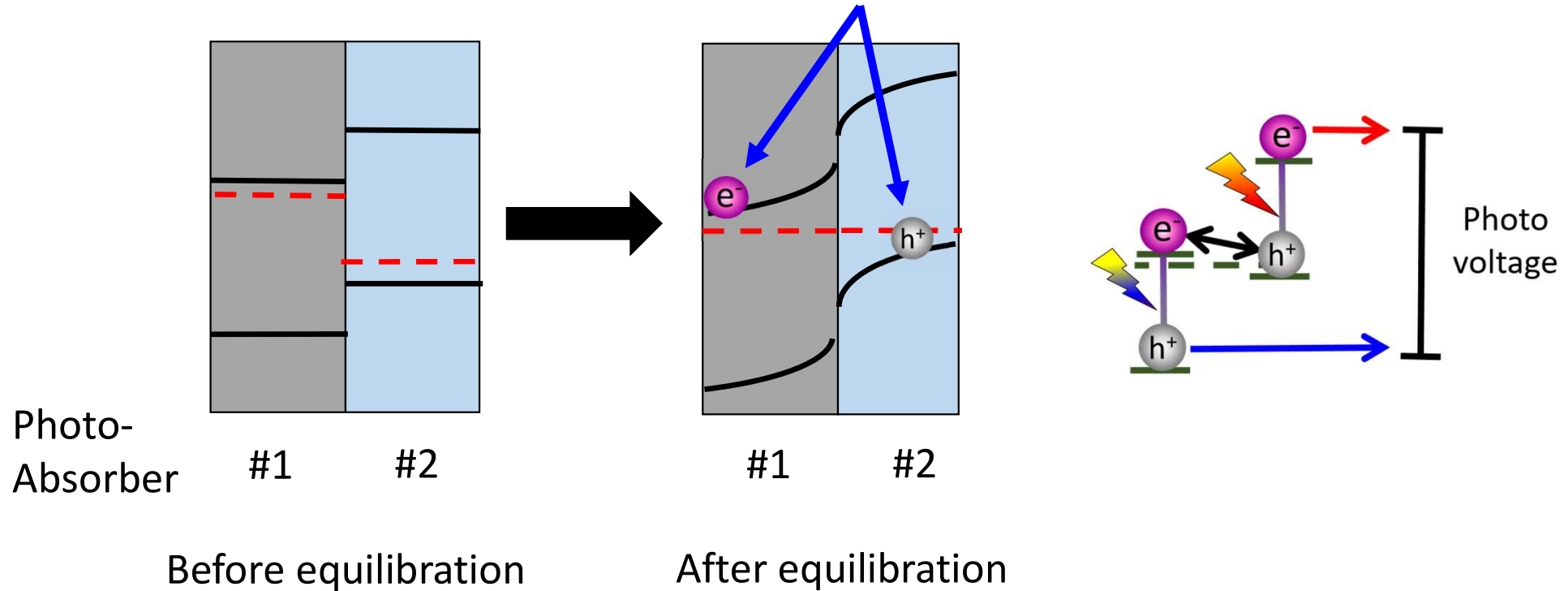
- If the band positions are not aligned at the interface we have an offset.
- This interface give us problems due to Fermi level alignment



# The interface between photoabsorbers

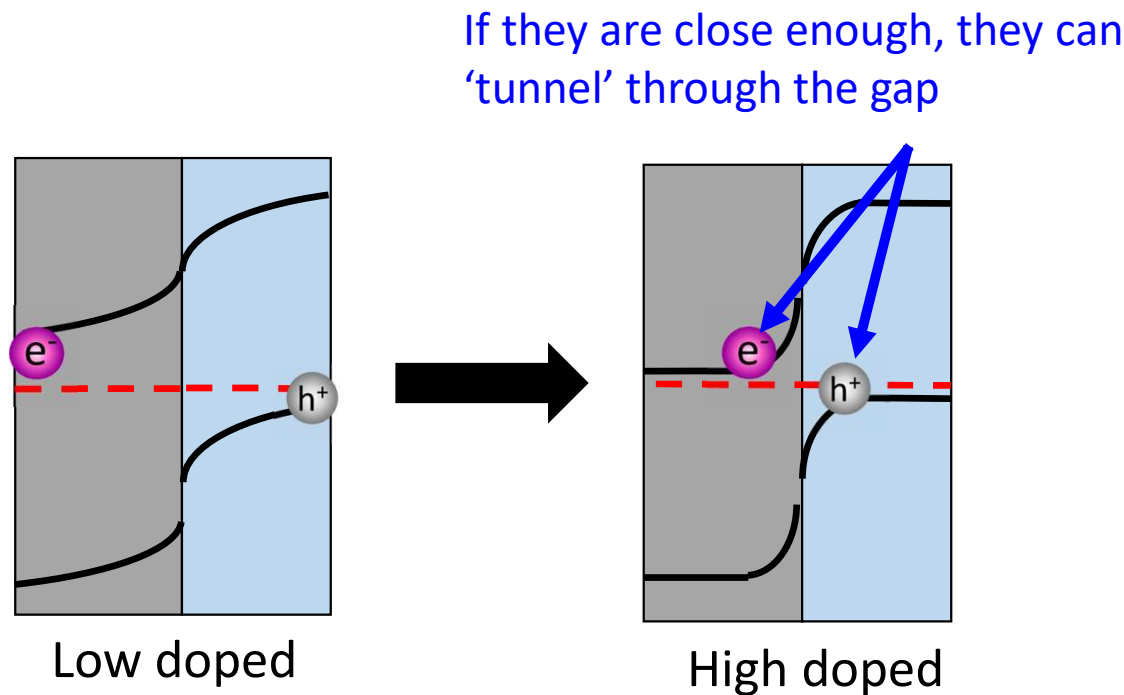
- There is a p-n junction at the interface going the wrong way

These are too far away to recombine. Thus they will recombine with carriers in their own semiconductor



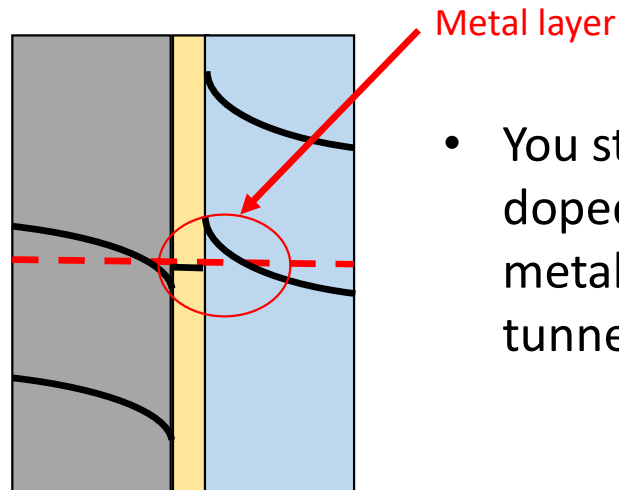
# Tunnel Junctions for Tandems

- Tunneling allows us to 'ohmically connect' two photoabsorbers
- Highly doping both sides provides a very short depletion layer



# Other tunneling approaches

- A thin metal layer can serve as a tunnel junction



Kind of highly doped

- You still need relatively highly doped photoabsorbers, but a metal in the middle cuts your tunneling distance by ~50%

Tunneling probability

$$T_t = \exp \left[ -\frac{4}{3} \Delta x \sqrt{\frac{2m_{eff} q \Phi_b}{\hbar^2}} \right]$$

- Metal work function should be close to correct position, but not essential.

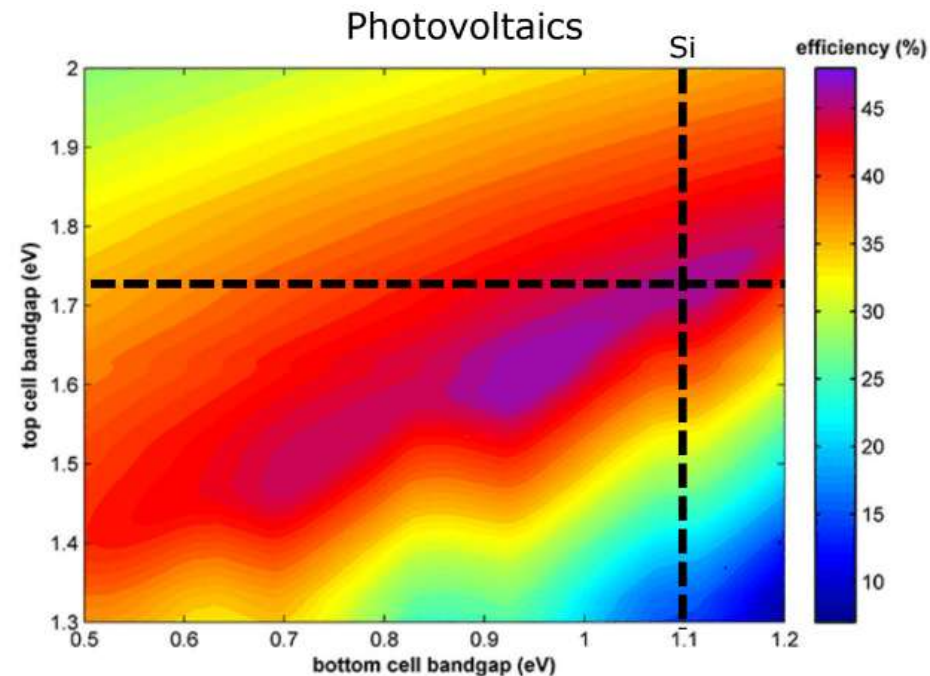
# Practical tandem materials to match Si

## Perovskite solar cells ( $\text{CH}_3\text{NH}_2\text{PbI}_x\text{Br}_y$ )

- Band gaps:  
 $\text{CH}_3\text{NH}_2\text{PbI} = 1.5 \text{ eV}$   
 $\text{CH}_3\text{NH}_2\text{PbBr} = 2.2 \text{ eV}$
- Mixtures photo-segregate

## Selenium solar cells

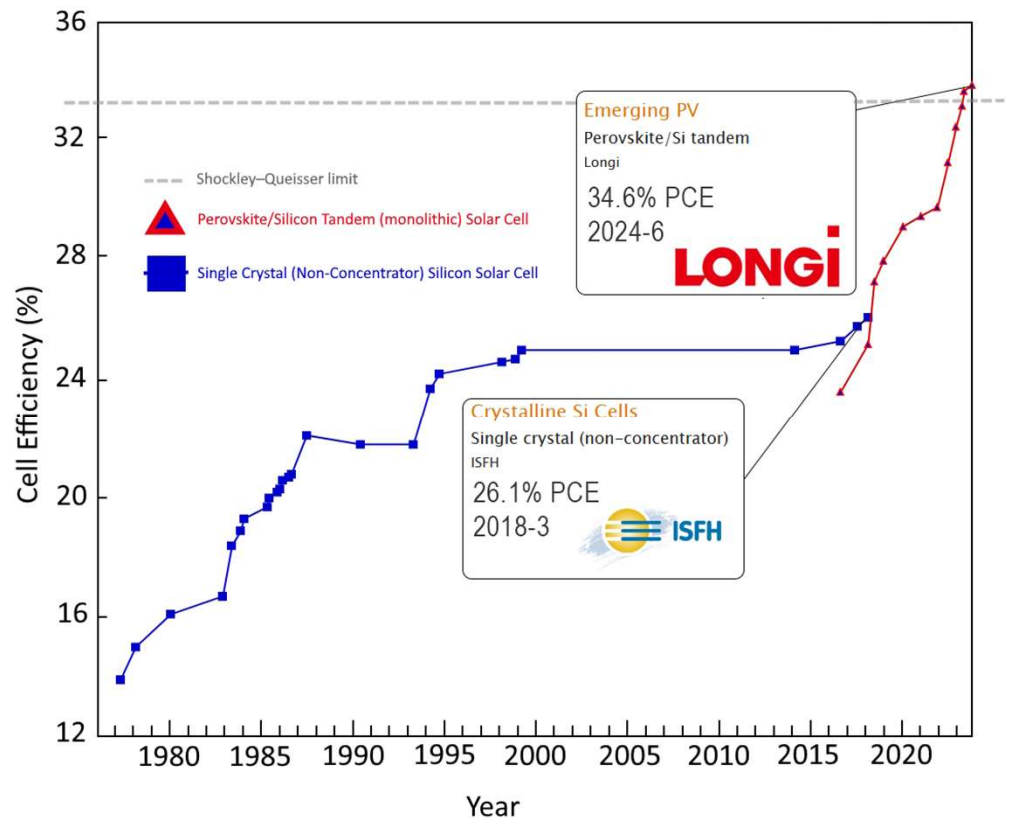
- Bandgap = 2 eV
- Low melting temperature ( $\sim 200 \text{ C}$ )
- First ever solar cell (1883)





# Optimal band gaps for tandem devices

- Longi has the record for 2-photon absorbers tandem at 34.85% (April 2025)
- The use a perovskite/Si Tandem cell
- Longi is a company, not a university/research-facility
- Their commercial devices operate at 30.1% efficiency



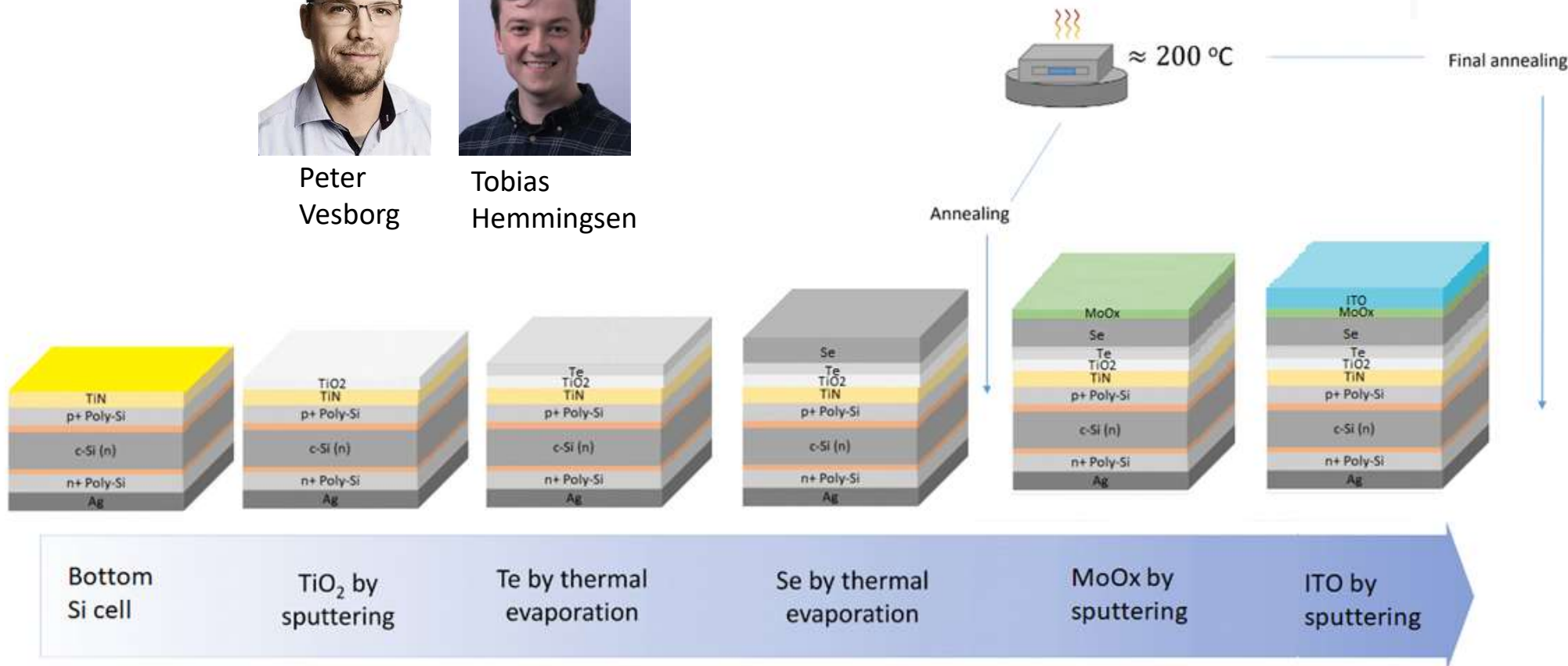
# Development of Se-Si at DTU



Peter  
Vesborg



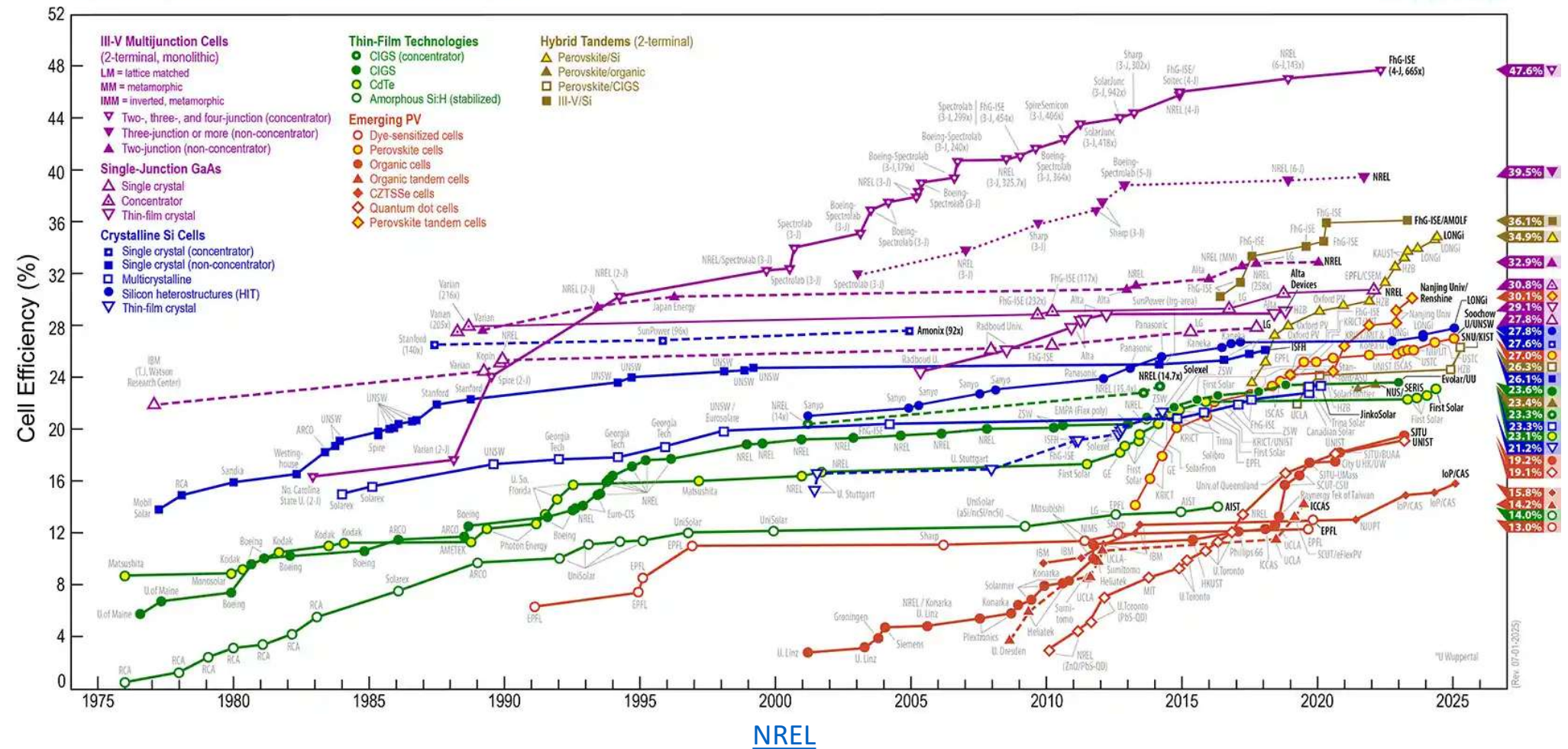
Tobias  
Hemmingsen



# Certified record solar cell efficiencies



## Best Research-Cell Efficiencies



# Lecture - Learning Objectives



*At the end of this lecture you should:*

- Understand the following junctions:
  - P-n junction
  - p-i-n junction
  - Hetero-junction
  - Tunnel junction
  - Schottkey Barrier
- Understand how to apply various junctions to make an overall solar cell device.
- Understand basics of tandem solar cells
- Understand bi-facial solar cells

# Exercises

- If you have a p-n junction Si with the n-type having a donor density of  $1 \times 10^{19} \text{ cm}^{-3}$  and the p-type donor density of  $2 \times 10^{16} \text{ cm}^{-3}$  plot the band bending at the p-n junction. Just do a rough plot showing the depletion layer and bias voltage.
- If we take Si with a  $\text{SiO}_2$  tunnel junction, how thick can the  $\text{SiO}_2$  be to still allow  $100 \text{ mA/cm}^2$  to pass through.

*(Assume 25C and the barrier height ( $q\theta_b$ ) for Si- $\text{SiO}_2$  is 4.35 eV, and an  $m_{\text{eff}}=1.0$  (\*mass of electron in vacuum). All other properties for Si can be found online. )*